

A Broad-Specificity β -Glucosidase from a Wild Type of Yeast Isolate and Its Potential Use in Food Industry

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ABSTRACT

A novel intracellular β -glucosidase was isolated from a wild type of yeast *Trichosporon* sp.). Production of enzyme was stimulated with the highest level of production in medium containing 1 % lactose as a sole carbon source at pH 7 and 30°C for 48 hr. The enzyme preparation exhibits activity towards several β -glucosidic substrates, indicating that, it has broad- specificity, furthermore the activity towards lactose hydrolysis was the highest as compared with other substrates. Also, it was able to catalyze transglycosylation. Therefore it was used for milk lactose hydrolysis as well as oligosaccharides production. It has, thus potential use in food industry.

Keywords: β -glucosidase, lactose hydrolysis, transglycosylation.

INTRODUCTION

The β -glucosidases catalyze the hydrolysis of various compounds with, β -D-glucosidic linkages (Woodward & Wiseman, 1982, Saha & Bothast, 1996) as well as the synthesis of oligosaccharides by reversal of their hydrolytic action (Hansson & Adlercreutz, 2001, Saloheimo *et al.*, 2002, Ishii-Karakasa, 2003). Accordingly, these enzymes have been widely utilized in the food industries as a tool for the hydrolysis of glucosidic linkages in various food materials, such as increase in starch recovery from potatoes, extraction of essential oils and the extraction of green tea components. In the flavour industry, β -glucosidases are key enzymes in the enzymatic release of aromatic compounds from glucosidic precursors present in fruits and fermenting products (Shoseyov *et al.*, 1990, Gueguen *et al.*, 1996). The β -glucosidases, are generally categorized as enzymes which hydrolyse β -(1-4) glucosidic linkage, have a wide variety of enzymatic properties, depending upon the origin and conditions under which the organism was grown (Han & Srinivasan, 1969). They are divided into three groups on the basis of substrate specificity (I) aryl- β -glucosidases, which have a strong affinity for aryl- β -glucosides (II) cellobiases, which hydrolyze only oligosaccharides (III) broad-specificity β -glucosidases, which exhibit activity on many substrate types and are the most commonly observed β -glucosidases (Rojas *et al.*, 1995). The β -glucosidase from the yeast (*Trichosporon* sp) is broad specificity type since it can hydrolyze many

types of β -diglycosides while lactose is the second substrate after cellobiose, thus it can be called enzyme with β -glucosidase and β -galactosidase activity as described previously by Nakkharat & Haltrich, (2006). They are isolated and characterized as enzymes from fungus *Talaromyces thermophilus* CBS 236.58 followed this phenomenon. Many β -glucosidases with very broad specificity have been isolated from many fungi (Kwon *et al.*, 1992, Copa-patino & Broda, 1994, Gueguen *et al.*, 1995, Pitson *et al.*, 1997). According to the classification of β -glucosidases based on substrate specificity most family 1 enzymes also show significant β -galactosidase activity (Bhatia *et al.*, 2000). The β -galactosidases catalyse the hydrolysis of lactose, in addition, they were shown to catalyse transgalactosylation reaction. Both the hydrolase and transferase activity of β -galactosidases have recently attracted interest because of possible applications in food industry, such as production of low lactose milk and synthesis of oligosaccharides (Cruz *et al.*, 1999, Karasov *et al.*, 2002, Vasiljvic & Jelen, 2003, Nakkharat & haltrich, 2006). Hydrolysis of lactose by β -galactosidases alleviates lactose maldigestion problems such as abdominal pain, flatulence or diarrhea which may result from the fermentation of undigested lactose by colonic bacteria with production of H₂, CH₄, CO₂ and short-chain organic acids (Rings *et al.*, 1994, Fooks *et al.*, 1999, Matioli *et al.*, 2001). Also, the hydrolysis of lactose is industrial beneficial to overcome lactose crystallization in condensed milk and ice cream (Karasov *et al.*

2002). Oligosaccharides synthesized by enzymes are mainly used as food additives in beverage, infants milk powder, yoghurts, chewing gums, and in the manufacture of candy, pastry, bread and jams because of their heat stability for modification of the colonic microflora toward a healthy balance by increasing the gut bifidobacteria and lactobacilli at the expense of *clostridia*, *proteolytic* and *Escherichia coli* bacteroids. This change in the intestinal flora composition has been proposed to be responsible for the decrease of putrefactive products in the feces, for a lower blood cholesterol content, higher Ca^{2+} absorption, a smaller loss of bone tissue in ovariectomized rats and a lower incidence of colon cancer (Cruz *et al.*, 1999, Fooks *et al.*, 1999, Sako *et al.*, 1999, Boon *et al.*, 2000, Van Laere *et al.*, 2000, Albayrak & Yang, 2002, Choi *et al.*, 2003) This work presents for the first time a novel enzyme with β -glucosidase and β -galactosidase activity produced by a newly isolated yeast strain and demonstrates its potential use in the production of low lactose milk and synthesis of oligosaccharides.

MATERIALS AND METHODS

The yeast strain and culture conditions

The yeast strain (*Trichosporon* sp.) used in the present study was isolated and identified at the Institute of Microbiology and Wine Research, Gutenberg University, Mainz, Germany. The medium used for optimization of β -glucosidase production was as follows: Twenty grams of Bacto yeast nitrogen base (Difco) with 1% (w/v) carbon source were dissolved in 1L of 50 mM buffer solution under studying pH value, then media solutions were filter sterilized. Erlenmeyer flasks (200-ml) containing 50 ml of medium was inoculated and incubated at test temperature on a rotary shaker (200 rpm) for 24 hr.

Production of β -glucosidase

The strain was grown in 5-liter Erlenmeyer flasks each containing 2000 ml of the medium which was optimized (20g Bacto yeast nitrogen base, 1% (w/v) lactose at pH 7 and 30°C for 48 hr, modified medium of Difco Bacto yeast nitrogen base). Cells were harvested by centrifugation 2000Xg, 15 min, washed twice with 50 mM sodium citrate buffer pH 5 and centrifuged again. The resultant paste was stored at -18°C until used.

Preparation of β -glucosidase

The cell paste was suspended in appropriate volume of 50 mM sodium citrate buffer pH 5 and homogenized with one volume of glass beads (diameter 0.5 mm). Cells were disrupted by shaking in vibrating homogenizer (Vibrogen Zellmühle, E. Bühler, Tübingen, Germany.) at 4°C for 6 min as described by Mireau *et al.* (2000). Debris was removed by centrifugation at 20000X g for 30 min, supernatant was desalted by dialysis against distilled water and lyophilized. For partial purification, the lyophilized preparation was dissolved in 20 mM Tris/HCl buffer, pH 7.6 and loaded onto a column (containing Q Sepharose Fast Flow, Pharmacia) which was previously equilibrated with 20 mM Tris/HCl buffer, pH 7.6. Elution was carried out with a linear gradient of 2M NaCl in the same buffer, at a flow rate of 1 ml/min. Active fractions were combined, desalted, lyophilized and used as enzyme preparation for further study.

Enzyme assay

One ml of 5 % lactose (0.139M) solution or other tested saccharides in 0.05M phosphate buffer pH 6.5, was mixed with the enzyme solution in total volume of 2.0 ml and incubated at 40°C. The reaction was stopped after 1 hr by heating the tubes in a boiling water bath for 5 min. One ml of the reaction mixture was deproteinized by adding 1.0 ml of 5 % $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$ and 1ml of 4.5% $\text{Ba}(\text{OH})_2$ and filtered. Glucose produced was estimated in 1ml of the filtrate by the glucose-oxidase method as cited by Cruz *et al.* (1981), with nitrophenylglucoside as substrate, the *p*-nitrophenol release at 40°C was monitored at 400 nm. In both cases an enzyme unit (U) was defined as the amount of enzyme necessary to liberate 1 μmol of the glucose or *p*-nitrophenol per min under the assay conditions (Makkar *et al.*, 1981)

Thermal stability

The enzyme solution (3 units) was incubated in 50 mM citrate buffer pH 5 at different temperatures ranging from 4°C to 80°C for 60 min., and then the remaining activities were determined using lactose as the substrate (Oikawa *et al.*, 1998)

pH stability

The enzyme solution (3 units) was treated with various buffers at 4°C for 3 hr., then the pH was readjusted to 6.5, then the remaining activities were

determined using lactose as the substrate. The following buffers (50 mM) were used: (a) glycine / HCl, pH 2-3, (b) citrate pH 4-5, (c) sodium phosphate pH 6-7, (b) Tris / HCl pH 8 and glycine / NaOH pH 9-10 (Oikawa *et al.*, 1998).

Milk sample

Milk sample was obtained from the farm of the Faculty of Agriculture, Kafr Elsheikh Univ. It had the following composition, 3.3% fat, 4.7% lactose, and 0.71 % ash. For whole milk, the sample was used without treatment. Skim milk sample was prepared by centrifuging the cold whole milk at 8000Xg for 30 min then the fat was removed.

Oligosaccharide synthesis

The basic reaction mixture containing enzyme preparation (2 units) and 1 ml of 250 mM lactose at tested pH values and temperatures was left for 16 hr, and then boiled for 5 min in water bath, and centrifuged at 1000X g for 3 min. The supernatants were filtered through syringe filter (0.45 μ m Minisart, Sartorius) and analyzed by HPLC equipped with an Aminex column HPX-42A, (300 mm x 7.8 mm). Twenty μ l of each sample were applied onto the column and eluted with deionized water at a flow rate of 0.6 ml/min. The reaction products were detected by refractive index and identified and quantitated by comparison with retention times of authentic appropriate sugars standards (Modified method of Choi *et al.*, 2003).

RESULTS AND DISCUSSION

Optimization of β -glucosidase production

The β -glucosidase properties are dependent upon the origin and conditions under which the organism was grown (Han & Srinivasan, 1969). But, the nutritional and environmental factors for β -glucosidase production by the strain under study are not known yet, thus the purpose of these growth studies was to determine the conditions that stimulate the enzyme production as follows:

Effect of growth temperature on enzyme production

The strain was cultivated at a temperature range of 10–50°C, for 24 hr in medium containing 1% (w/v) cellobiose as a carbon source at pH 7 to determine the optimal temperature for enzyme production, the results were summarized in Fig. (1). The yeast was able to grow and release β -glucosi-

dase at a wide range of temperature. The maximum enzyme yield (5.7U/ml) was produced at 30°C.

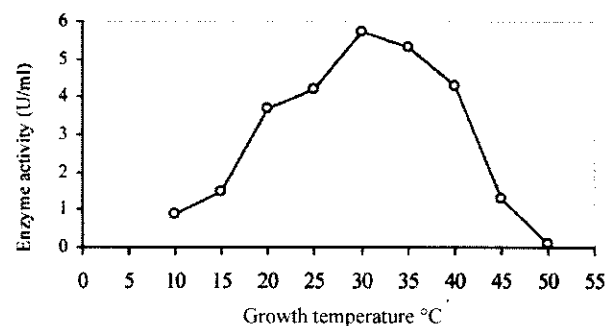


Fig. 1. Effect of cultivation temperature on enzyme production

Effect of initial pH of culture medium on enzyme production

The strain was cultured at various pH values from 3 to 10, for 24 hr in medium containing cellobiose as a carbon source at 30°C. Fig. (2). shows that the yeast was able to grow and produce the enzyme at a broad range of pH, while the highest enzyme yield (5.9 U/ml) was gained at pH 7.

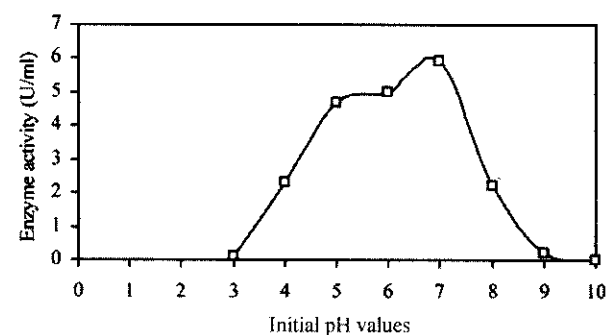


Fig. 2. Effect of medium pH on enzyme production

Effect of carbon source on enzyme production

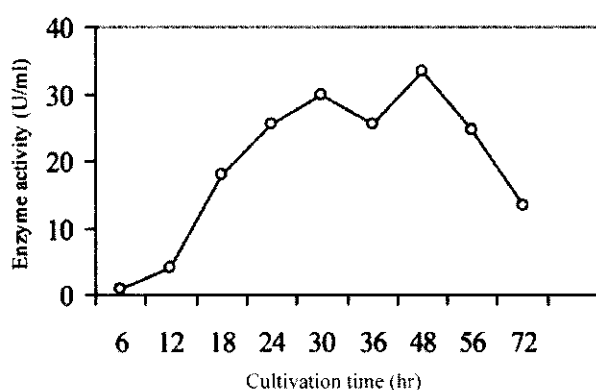
Various β -glucosidic and non- β -glucosidic carbohydrates were employed in equal concentrations of 1% (w/v) to test their ability for inducing the formation of β -glucosidase (Table 1). In general, the yeast can utilize all tested carbohydrates with diversity in the enzyme production. The formation of the highest amount of β -glucosidase was induced by lactose. It could be also observed that the enzyme formation is not dependent on the kind of glucosidic bond since cellobiose, maltose and isomaltose induced the formation of approximately the same enzyme yield. These results are in agreement with the observation of other workers (Skory *et al.*, 1996, Riou *et al.*, 1998) for optimization of β -glucosidase production from other organisms.

Table 1: Effect of various carbohydrates on enzyme production

Carbon source	β -glucosidase activity (U/ml)
Glucose	5.5
Salicin	5.3
Maltose	26.6
Isomaltose	26.1
Raffinose	2.1
Lactose	31.0
Cellobiose	26.8
Sucrose	6.1

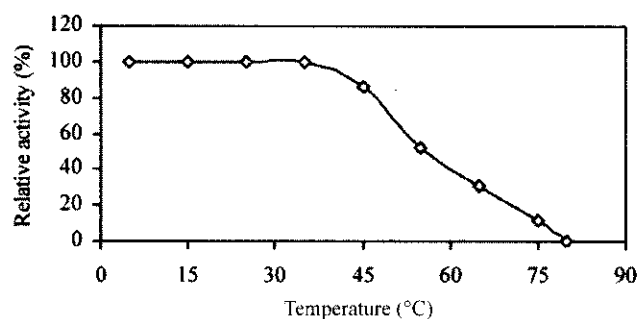
Time course of β -glucosidase production

The yeast strain was cultivated under the aforementioned optimal cultural conditions for 72hr. Fig. (3) shows that the enzyme activity (0.9U/ml) was detected after 6hr of incubation and increased gradually to reach its maximum (33.45U/ml) within 48 hr of the cultivation, and then started to decrease.

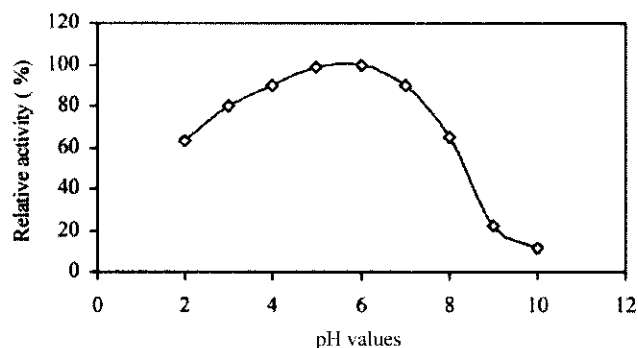
**Fig. (3). Effect of incubation periods on enzyme production****Thermal stability**

The results of thermal stability experiments results are shown in Fig. (4), which reveals that the enzyme preparation maintains its activity up

to 35°C and even at 40°C, it retains 86.1 % of its maximum activity at 45°C and drops rapidly, to be completely inactivated at 80°C.

**Fig. 4. Effects of temperature on the enzyme stability****The pH stability**

The enzyme preparation was quite stable at pH 5 to 6 Fig.(5). The activity gradually retarded above and below the mentioned values.

**Fig. 5. Effects of pH on the enzyme preparation stability****Substrate specificity of β -glucosidase preparation**

The enzyme preparation exhibited a broad specificity for hydrolyzing various carbohydrate types (Table 2). The rate of hydrolysis depended on the nature of aglycon moiety and the type of linkage. The highest hydrolyzing activity was observed with substrate having β -(1,4)-glucosidic linkages, such as

Table 2: Relative values rates of hydrolysis of various substrates by the enzyme preparation

Substrate	Type of linkage	Relative hydrolysis rate (%)
<i>p</i> -Nitrophenyl- β -D-glucopyranoside	β Glc	100
Sucrose	β , α (1,2)	00.0
Lactose	β (1,4) Gal	119.3
Salicin	β Glc	18.0
Raffinose	α (1,6), β (1,2)	3.2
Sophorose	β (1,2) Glc	28.7
Gentiobiose	β (1,6) Glc	79.0
Cellobiose	β (1,4) Glc	131
carboxymethylcellulose	Poly β (1,4) Glc	00.0
Avicel	Poly β (1,4) Glc	00.0

cellobiose and lactose. The enzyme could also hydrolyze the saccharides with β - (1,6) glucosidic linkages effectively. The enzyme, however, had very little (28.7, 3.2% and 0%) or no activity on sophorose, raffinose, and sucrose, respectively. On the other hand, activity against polysaccharides carboxymethylcellulose (CMC) and acid-swollen avicel was not found. From these results, it can be suggested that the enzyme belonged to class "Broad substrate specificity β -glucosidase enzyme" according to the classification system based on substrate specificity (Bhattachia *et al.*, 2002). The β -glucosidases with very broad specificity have been isolated from many bacteria, yeast and fungi (Han & Srinivasan, 1969, Wood & McCrae, 1982, Yan & Lin, 1997). The enzyme efficiency towards lactose hydrolysis paid our attention to do additional studies about its catalytic ability.

Hydrolysis of milk lactose

The milk is consumed in two preferred forms skim and whole milk containing normally about 4.7 % fat, thus the present study aimed to define the optimum conditions for hydrolysis of milk lactose in both forms. The main factors, which affect the enzyme activity include reaction pH, reaction temperature, substrate concentration, etc. Since the substrate concentration (lactose in raw milk) is approximately stable, the study was focused on the other mentioned factors.

Effect of temperature on hydrolysis of milk lactose by enzyme preparation

To investigate the effect of temperature on enzyme activity, the enzyme was incubated with

milk at temperatures ranged from 30 to 50°C. The results given in Table (3) reveal that, the reaction was normally affected by temperature and the maximum hydrolysis yield was found at 35°C with both milk forms under study, then a gradual fall in the hydrolysis yield was observed. In comparison in other yeasts (*Kluyveromyces fragilis* and *Kluyveromyces lactis*) regarding the enzymes used for lactose hydrolysis, they showed optimum temperature at 40°C (Matioli *et al.*, 2001, Genari *et al.*, 2003).

Effect of reaction pH on hydrolysis of milk lactose

The results in Table (4) indicate that the enzyme was capable to hydrolyze milk lactose at various tested pH values, while luckily the maximum activity was coincided with the pH value 6.5 of the natural milk. This property allows using the enzyme in dairy industry without modification of milk pH. These results are in agreement with that of Makkar *et al.* (1981) and Matioli *et al.* (2001), they reported that pH 6.5 was the optimum value for lactose hydrolysis by the enzymes from *Kluyveromyces fragilis* and *Lactobacillus bulgaricus*.

Time course of milk lactose hydrolysis.

Various enzyme concentrations from (2 U/ml to 5 U/ml) were used to hydrolyze the lactose in both skim and whole milk at above estimated optimum temperature and pH for different time periods from 1hr to 5hr. Study aimed to determine the minimum enzyme concentration which reduces the lactose content to be satisfactory for lactose intolerance at minimum time to prevent any change in

Table 3: Effect of temperature on hydrolysis of milk lactose

Temperature °C	Lactose conversion in whole milk %	Lactose conversion in skim milk %
30	35.90	36.85
35	38.23	40.32
40	30.00	32.13
45	26.33	28.66
50	21.61	23.51

Table 4: Effect of reaction pH on hydrolysis of milk lactose

pH values	Lactose conversion in whole milk %	Lactose conversion in skim milk %
5.5	27.00	27.35
6.0	27.80	29.11
6.5	29.19	31.91
7.0	26.04	27.43
7.5	21.11	24.22

milk properties during the treatment. The experimental results in Table (5) demonstrate that, in general, the lactose hydrolysis increased by increasing the enzyme concentration and extending the reaction time. The activity towards hydrolysis of skim milk was slightly higher than that of whole milk at all determinations. It is satisfactory for lactose intolerants to hydrolyze 90 % of lactose in a product contains about 50 g/l lactose (Hernandez & Asenjo, 1982). Therefore, based on the experimental results (Table 5) the whole and skim milk treated with amount of enzyme 5U/ml for 4 hr was satisfactory for lactose intolerants. On the other hand, skim and whole milk treated with 4 U/ml for 3hr was readily accepted from the industrial point of view according to the recommendation of Prenosil *et al.*, (1987). They indicated that hydrolysis of 75 % - 85% of lactose in milk containing 5 % lactose is industrially accepted. In comparison, with other enzymes from other organisms, the use of 3450 U/l β -galactosidase from *Kluyveromyces fragilis* caused a conversion of 70 % of lactose from substrate contain 5% (w/v) lactose at pH 6.5 and 40°C in 2 hours (Matioli *et al.*, 2003).

Oligosaccharides formation

Effect of temperature on the synthesis of oligosaccharides

To determine the optimum temperature of oligosaccharides formation, the reaction was carried

Table 5: Time course of milk -lactose hydrolysis

Time (hr)	Whole milk				Skim milk			
	Enzyme concentrations				Enzyme concentrations			
	2U/ml	3U/ml	4U/ml	5U/ml	2U/ml	3U/ml	4U/ml	5U/ml
1	18.00	23.33	28.10	30.12	19.19	22.91	29.08	31.90
2	37.44	46.16	56.54	65.67	38.00	47.51	57.90	66.60
3	45.16	57.74	71.80	84.90	45.80	58.42	71.90	86.12
4	52.88	69.32	87.42	92.27	53.00	70.31	88.61	93.05
5	60.61	70.90	90.15	96.10	60.90	78.00	90.11	95.23

Table 6: Effect of temperature on galactooligosaccharide synthesis

Temperature °C	Reaction products (mM)			
	Glucose	galactose	Lactose	Galactooligosaccharide
20	95.25	84.01	158.84	0.71
30	152.53	107.05	118.68	1.21
40	128.75	99.42	134.59	3.88
50	125.12	95.28	140.64	1.01
60	47.14	43.01	204.81	0.00

Reaction conditions: substrate concentration 250mM; Enzyme concentration, 2U/ml and pH 7 for 16hr.

out at temperature range of 20-60°C. The results in Table (6) show that the formation of oligosaccharides increased with the increase of temperature to reach its maximum at 40°C and then started to decline. These results are in agreement with the observation of Monsan & Paul (1995) and Vasiljevic & Jelen (2003) they reported that the amount and the rate of oligosaccharides formation were significantly affected by the reaction temperature and pH.

Effect of pH on oligosaccharides formation

To define the optimum pH for oligosaccharides synthesis, the reaction pH was adjusted at range of 4 – 9. The results inTable (7) showed that, the highest amount of oligosaccharides was formed at pH 8. It could be also noted that, the optimum temperature and pH for lactose hydrolysis by the same enzyme were 35°C and 6.5 (Tables 3 and 4) whereas in case of oligosaccharides synthesis they were shifted to be 40°C and pH 8, this observation was previously mentioned by Cruz *et al.*, (1999).

Effect of substrate concentration on oligosaccharides formation.

In order to test the influence of lactose concentration on synthesis of the galactooligosaccharides, enzyme preparation (2U/ml) was incubated with various lactose concentrations. The results given in Table (8) and Fig. (6) reveal that, in general this novel enzyme was able to catalyze transgalactosylation reaction with lactose as a substrate.

Table 7: Effect of reaction pH on galactooligosaccharide synthesis

Reaction pH	Reaction products (mM)			
	glucose	galactose	lactose	galactooligosaccharide
4	70.28	68.01	179.85	0.04
5	116.00	99.05	138.89	1.80
6	124.76	120.92	122.66	3.00
7	125.11	118.28	121.59	4.01
8	121.04	110.91	125.81	4.68
9	49.99	46.85	199.82	1.62

Reaction conditions: Substrate concentration 250mM; Enzyme concentration, 2U/ml and Temperature, 40°C for 16hr. The following buffers (50 mM) were used: (a) Citrate pH 4 - 5, (b) Sodium phosphate pH 6 - 7, (c) Tris / HCl pH 8 and glycine / NaOH pH 9.

Table 8: Effect of substrate concentration on galactooligosaccharide synthesis

Substrate concentration (mM)	Reaction products (mM)			
	glucose	galactose	lactose	galactooligosaccharide
100	99.87	98.92	0.0	0.0
200	166.54	131.46	48.58	2.44
300	224.29	158.80	98.89	6.84
400	264.95	179.54	158.61	13.82

Reaction conditions: pH 8, Enzyme concentration, 2U/ml and Temperature, 40°C for 16hr.

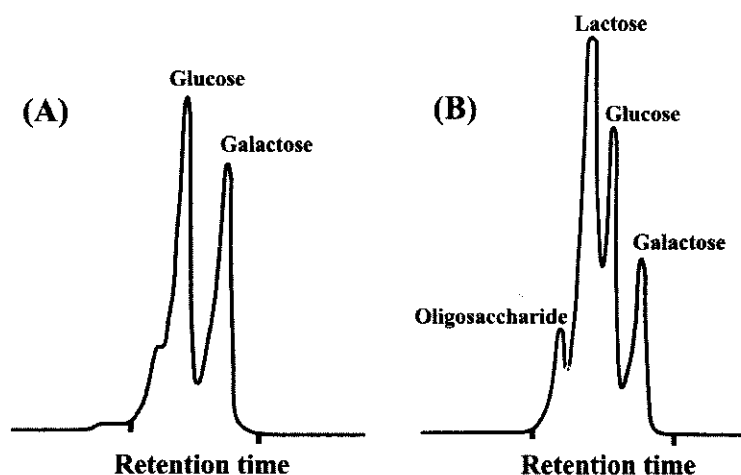


Fig . 6 : HPLC Chromatogram of lactose hydrolysis at low lactose concentration (A) oligosaccharide formation at high lactose concentration (B)

The formation of oligosaccharides was absolutely dependent on lactose concentration, at lactose concentration lower than 200 mM, the hydrolysis reaction was dominated, but above the mentioned concentration, the reaction was shifted towards the formation of oligosaccharides. This may be attributed to β -galactosyl groups having a higher ability of attaching to lactose than water as an acceptor at increasing lactose concentration (Iwasaki *et al.* 1996). The formation of oligosaccharides was gradually increased with the increasing of lactose

concentration to reach its maximum at 400 mM. These observations are in accordance with the findings of Boon *et al.*, (2000) and Vasilijevic & Jelen, (2003) who reported that, initial lactose concentration is the most significant factor affecting the oligosaccharides synthesis. Data also reveal that, the yield of oligosaccharides obtained here was lower than that obtained by other workers from other microbial enzymes (Rabiu *et al.*, 2001, Albayrak & yang, 2002, Choi *et al.*, 2003). This may be attributed to, the reaction of oligosaccharides synthe-

sis activated by the enzyme preparation under study still need more optimization by studying the other parameters affecting the oligosaccharides formations such as enzyme concentration, reaction time, activators, inhibitors, etc. However, the previous results cleared for the first time, this novel enzyme is capable of producing oligosaccharides by transgalactosylation, which have high biological value. Also the results indicated that, the produced yield of oligosaccharides was able to be enhanced by optimizing the factors, which affect the reaction such as temperature, pH and substrate concentration.

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إنتاج إنزيم البيتا جلوكوسيديز واسع النشاط من سلالة بيرية للخميرة وامكانية استخدامه في مجال الصناعات الغذائية

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أمكن عزل إنزيم البيتا جلوكوسيديز لأول مرة من خميرة الـ *Trichosporon* sp عند نموها في ظروف مثالية في بيئة رقم الأس الهيدروجيني لها ٧ و تحتوي علي ١٪ لاكتوز كمصدر وحيد للكربون لمدة ٤٨ ساعة علي درجة حرارة ٣٠ درجة مئوية. و وجد أن مستحضر الإنزيم له نشاط تجاه العديد من المركبات الجليكوسيدية ذات الرابطة بيتا وكان أعلي نشاط له تجاه سكر اللاكتوز لذا فإنه استخدم لخفض نسبة سكر اللاكتوز في اللبن كما استخدم لتخليق سكريات الأوليجو ذات القيمة الحيوية العالية من اللاكتوز لذا فهو ذو قيمة تطبيقية عالية في مجال الصناعات الغذائية.