EXTRACTABILITY AND AVAILABILITY OF SOME NUTRIENTS IN CONTAMINATED SOILS OF ASSIUT.*

Shaymaa, M.K.

Assiut University, Faculty of Agriculture, Soil and Water Department

Abstract: Various extraction procedures were employed for measuring extractable concentrations of potential micronutrients and toxic elements in polluted soils which were irrigated by sewage waste waters for a long time (>45 years).

The extractability of Fe, Zn, Mn, Cu, Cd, Ni and Cr in six contaminated soils near Assiut city using eight different extraction procedures, (HCl + H2SO4, DTPA, AB-DTPA, EDTA, EDTA-Aac, EDTA-Ammonium carbonate and EDTA-Ammonium Caitrate) as well as total element contents were studied. Extractability of metals from soils

samples varied depending on types of metals and extractants used.

Our results emphasized that DTPA extractant may not be a good extractant for heavy metals, and its use for extraction of heavy metals from contaminated soils is questionable. EDTA extraction procedures were not specific for all micronutrients and heavy metal extractions. These results lead to the general conclusion that no one extraction procedures is suitable for testing the status of all metals in all soils after all crops. Precautions should be taken when deciding which extraction solution should be used.

key words: extractability, availability, nutrients, contaminated soils.

Introduction

Heavy metals pollution of soils is increasingly becoming a problem with global the development of industry, irrigation of using wastewater and the application of sewage sludge even if it is relatively localized at present. The soilplant system is the fundamental constructive unit ofthe geosphere biosphere. and Therefore, heavy metals pollution of soil has an important influence not only on the yield

and quality of crops, but also on the quality of atmospheric and aquatic environment, and even on the health of human beings via food chains. Trace metals in soils may exist in different chemical forms or ways of binding. In unpolluted soils or sediments trace metals are mainly bound to silicates and primary minerals forming relatively immobile species, whereas in polluted ones trace metals are generally more mobile and bound to other soil or sediment phases.

^{*}Submitted to the 2nd Conference Of Junior Scientists, Fac. of Agric, Assiut Univ. May., 6,2008.

In environmental studies the determination of the different ways of binding gives more information trace metal on mobility, as well as on their availability toxicity or comparison with the total element content. However, the determination of the different ways of binding is difficult and impossible. Different approaches are used for soil and sediment analysis. Many of them focused on pollutant desorption from the solid phase, others are focused on the pollutant adsorption from a solution by the phase. Among approaches based on desorption; leaching procedures are the most widely accepted and used.

Extraction procedures bv means of a single extractant are widely used in soil science. These procedures are designed to dissolve a phase whose element content is correlated with the availability of the element to the plants. This approach is well established for major elements and nutrients and it is commonly applied in studies of fertility and quality of crops, for predicting the uptake of essential elements, for diagnosis of deficiency or excess of one element in a soil, in studies of the physico-chemical behavior of elements in soils and for survey purposes. To a lesser extent they are applied elements considered as pollutants metals. as heavy The application of extraction

procedures polluted to naturally contaminated soils is mainly focused to ascertain the potential availability mobility of pollutants which is related to soil-plant transfer of pollutants and to study migration in a soil profile which usually connected with groundwater problems.

During the last decades several extraction procedures for extracting heavy metals in soils been developed have and modified. In this respect, two must groups of tests considered: the single reagent extraction test, (one extraction solution and one soil sample), and the sequential extraction procedures, (several extraction solutions are used sequentially to the same sample) although this last type of extraction is still in development for soils. Both typesof extraction are applied using not only different extracting different schemes also but laboratory conditions. This lead to the use of a great deal of procedures. extraction transfer of heavy metals from soils to plants is dependent on three factors: the total amount of potentially available elements (quantity factor), the activity as well as the ionic ratios of elements in the soil solution (intensity factor), and the rate of element transfer from solid to liquid phases and to plant roots (reaction kinetics) (Bruemmer et al, 1986). Because correlations

were found between the soluble heavy metals concentrations in the soil and the heavy metals concentration in plants, several countries passed legislation establishing quality standards based on soluble heavy metals concentration in the soil. These concentrations must be reduced below maximum threshold levels avoid reduction in plant growth or nutritional quality. Moreover, the leaching of metals into the ground water must be minimized.

This research was designed to determine the extractability of some micro- and heavy metals in soils of six villages irrigated with sewage waste waters for along period (> 45 years) near Assiut different city using single extraction methods as well as total. In the other part of this study, the metals uptake by corn and wheat plants is investigated and correlated with the nutrient contents in the polluted soils. The obtained data when applying these testes are used for decision makers in topics such as land use of soil or in countermeasures application.

Materials and Methods

Study area:

Madabegh county wastewater management system uses aeration, short- term storage and slow rate irrigation. Over 90% of the effluent originates from

domestic sources, most of which nonorganic chemical are manufacturing and chemical related. An average of 80×10^3 m³/d of wastewater is conveyed to the system by collection network of gravity sewers and stations. pumping The wastewaters have been applied to the fields. through flood irrigation system, to a big area of 8000 - 10000 feddan (1 Feddan = 1.038 acre) for a long time (40-45 years) without any other pretreatment.

Soils and Water samples:

Along the irrigation cannel six locations (Madabegh, Mankabad, Bani Hussien, Ellwan, Ghalib, Gahdam) represented the diversity in soil properties of the studied area were chosen. Each location was represented by three from different sites which representative composite samples were collected. Eighteen soil profiles (6 locations, of three sites each) were made samples from surface (0-25 cm) and subsurface (25-50cm) layers were collected. The collected soil samples were air dried, crushed and passed through a 2mm sieve for analysis. Soil and kept certain analysis for soil properties (particle size distribution, E.C., pH, O.M.%, Ca, Mg, N, P, K and heavy metals were measured in each sample) of both layers summarized in Tables (1 and 2).

Table(1): Particle size distribution and texture grade of the studied soils.

Location	Sites	Depth (cm)	Clay	Silt	Sand	Texture
(Village)	(Profile)	Depar (cm)		%		grade@
	1	0-25	2.65	4.34	93.00	S
	1	25-50	5.70	2.40	91.90	S
	_	0-25	5.87	3.80	90.33	S
El Madabegh	2	25-50	3.07	3.00	93.93	S
	2	0-25	8.85	5.00	86.14	S
	3	25-50	5.85	2.25	91.90	S
	1	0-25	6.79	1.97	91.24	L. S.
	1	25-50	9.82	13.30	76.88	L. S.
	2	0-25	14.27	15.55	70.18	S. L.
Mankabad	2	25-50	8.95	20.66	70.40	S. L.
	3	0-25	14.05	10.50	75.44	S. L.
		25-50	11.71	14.67	73.62	S. L.
	1	0-25	3.85	13.31	82.84	L. S
	1	25-50	13.47	26.70	59.83	S. L.
Ellwan	2	0-25	12.22	15.06	72.72	S. L.
Eilwan	2	25-50	8.98	21.23	69.79	S. L.
	3	0-25	7.97	18.90	73.12	S. L.
	3	25-50	14.20	15.10	70.70	S. L.
	1	0-25	17.94	11.08	70.98	S. L.
		25-50	10.48	24.12	65.40	S. L.
Bani Hussien	2	0-25	8.01	23.84	68.15	S. L.
		25-50	9.47	15.72	74.82	S. L.
	3	0-25	5.10	13.11	81.80	L. S.
	3	25-50	18.00	5.54	76.46	S. L.
	1	0-25	6.34	1.97	91.69	S
		25-50	5.47	11.40	83.12	L. S.
75 COL 175	2	0-25	7.30	13.77	78.93	L.S.
Bani Ghalib		25-50	9.18	8.44	82.38	L. S.
	3	0-25	5.30	23.28	71.42	S. L.
	, [25-50	13.29	17.46	69.25	S L.
	1	0-25	9.00	4.03	86.97	L. S.
		25-50	16.18	25.58	58.24	\$. L.
Gahdam	2	0-25	7.54	5.87	86.59	L. S.
		25-50	4.70	7.42	87.88	L. S
	3	0-25	9.20	20.94	69.86	S. L.
		25-50	15.61	26.44	57.95	S. L.

[@] S, L.S, and S.L. mean Sandy, Loamy sand, and sandy loam, respectively.

Table(2): Some chemical properties of the soils of the studied soils

Village	Depth	EC	CEC	pН	9	6	m	eq / 100		P
Village	Deput	1:1	Cmol ⁺ /kg	1:1	O.M	N	K⁺	Ca⁺⁺	Mg [↔]	ppm
El Madabeg	0-25	1.0	0.82	7.10	3.00	0.07	27.50	15.73	8.80	27.33
El Madabeg	25-50	1.8	0.46	7.60	1.40	0.08	28.02	21.73	5.73	22.67
Mankabad	0-25	1.5	2.76	7.81	2.10	0.08	25.40	36.00	4.00	25.91
IVIAIIKADAU	25-50	3.5	2.40	8.05	1.50	0.09	28.72	36.67	8.40	20.38
Ellwan	0-25	0.5	4.74	7.56	3.30	0.09	27.50	39.33	7.33	25.75
Eliwan	25-50	1.4	4.15	7.76	1.80	0.08	27.15	39.33	15.33	20.46
Bani Hussien	0-25	1.3	4.65	7.59	2.40	0.09	27.67	39.33	10.67	24.87
	25-50	2.4	5.55	7.75	1.90	0.08	28.19	33.33	16.00	20.63
Bani Ghalib	0-25	2.5	3.79	7.96	2.90	0.08	26.27	21.33	19.07	25.36
Dail Gliallo	25-50	4.9	3.34	8.03	1.50	0.08	26.97	23.33	8.67	21.90
Gahdam	0-25	2.0	3.03	7.71	2.80	0.08	27.15	35.33	14.67	25.01
Candani	25-50	3.3	2.80	7.86	2.30	0.08	27.93	34.13	8.53	22.35

Heavy metals extractability:

In the first part of this study, the extractability of heavy metals of different single-extraction procedures was investigated. Nine different single-extraction procedures of variable pH (ranged between 4.5 and 7.5), strength and concentration of the active ingredients, and the operations of the procedure (soil: extractant ratio, shaking and

equilibrium time) were selected. The detailed operations were identical to the references in Table (3). In the same time, total concentrations of heavy metals in the soils of the studied area were determined after digestion using the three-acids mixture (HClO₄ + H₂SO₄ + HNO₃) according to Chapman and Pratt (1961).

Table(3): Heavy metals extraction procedures used in the study.

Extractant	pН	Time	Ratio	References
1) HCl + H ₂ SO ₄ (0.05 N HCl + 0.025 N H ₂ SO ₄)	1.58	15 min	1:4	El-Koumey et al., (1997) Wear and Evans (1968)
2) DTPA (0.1 M TEA+ 0.01 M CaCl2+ 0.005 M DTPA)	7.3	2 h	1:2	El-Koumey et al.,(1997)
3) AB-DTPA (Ammonium bicarbonate DTPA, 0.005 M DTPA+ 1 M NH ₄ HCO ₃)	7.6	15 min	1:2	Zhu and Alva (1993)
4) EDTA (0.05 M EDTA)	7	1h	1:10	Feng et al., (2005)
5) EDTA-Ammonium acetate (0.007 M EDTA + 1N ammonium acetate)	7	1 h	1:10	Trierweiler and Lindsay (1969)
6) EDTA-AAAc (EDTA- Ammonium acetate with acetic acid), (0.5 M NH ₄ OAc + 0.5 M HOAc + 0.02 M Na ₂ -EDTA)	4.65	1 h	1:10	Abd El-Haleem et al., (2002)
7) EDTA Ammonium carbonate (0.01 M EDTA + 1M (NH ₄) ₂ CO ₃)	8.6	1 h	1:10	Trierweiler and Lindsay (1969)
8) EDTA-Ammonium Citrate (0.01M EDTA+ 1 M NH ₄ - Citrate)	4.8	1 h	1:10	El-Koumey et al., (1997)

Pot experiments:

Two pot experiments were conducted in the greenhouse at the Faculty of Agriculture. Assiut University at Assiut. Soil samples. 5 Kg each, were packed in plastic pots of 15X17.5 cm, irrigated and left for one weak for drying. Com (Nifertete 3) seeds (10 seeds/ pot) were sown in each pot and thinned to 4 seedlings per pot after 10 days. Plants were watered to field capacity. When seedlings grow up to the first true-leaf (2 weeks), the plants were supplied with sufficient amounts of NPK at a rates of 0.16 gm N/ pot, 0.33 gm P/ pot and 0.28 gm K/ pot, in the forms of NH₄NO₃, KH₂PO₄ and K₂SO₄. Water loss was compensated for by daily addition of distilled water. Thirteen weeks (75 days) after planting, (aerial portion) were harvested, and soil samples were collected from each pot. To study the effects of the following crop on the extractability of polluted metals wheat plants were grown in the same pots. Ten seeds of wheat were sown in each pot (Giza 168). After 20 days from planting, wheat plants were thinned to 6 plants/pot. Pots were fertilized with N. P and K as mentioned in the first experiment. Plant samples were collected after 90 days and soil samples were collected from each pot. Corn and were freshly weighed. washed with distilled water, oven dried at 70° C for 24 hours to constant weight, dry weighed, ground to pass through a 20 mesh sieve and stored for analysis. Plant

dry materials were digested using $H_2SO_4 + H_2O_2$ acid mixture as described by Parkinson and Allen, (1975) and prepared for heavy metals and micronutrients determination. Soil samples were air-dried, crashed to pass from 2 mm sieve and stored for metal extractions later on.

Results and Discussion

Part 1:

Total metal contents in soils:

Table (4) and Figures (1-2) show the variation in total amounts of Fe, Mn, Zn, Cu, Cd, Ni, Pb and Cr in soils of the six locations (villages) of the study area irrigated by untreated swage wastewater for 40-45 years. Total amount of Fe, Zn, Mn, Cu, Cd, Cd, Pb, Ni and Cr in the different locations of the studied area were decreased in surface and subsurface layers in the following order: Fe > Mn > Zn > Cu > Ni > Pb > Cr > Cd. The data point to a decreasing order between the beginnings of the irrigation canal (Madabegh) toward its end at (Gahdam) along a distance of about 20 km. This may be attributed to the probable variations in the settlement processes downward the stream of the irrigation cannel. Most of the solid materials carried out with the sewage wastes present in irrigation water settled at the beginning of the in igation cannel and gradually decrease distance toward the end of the irrigation cannel. Similar results were obtained by El Sokkary and Sharaf (1996) and Abdel Salam (2002).

Table(4): Total amounts of micronutrients and heavy metals (ppm) in soils of the studied area.

Locations	Site	Fe	Zn	Mn	Cu	Çd	Pb	Ni	Cr
(Villages)	·			S	urface lay	rer	, .		
	1	5904.9	1485.2	382.7	551.8	2.3	75.5	46.2	15.1
geq	2	4314.0	1649.6	501.5	213.1	1.5	39.3	40.6	6.6
Madabeg	3	4415.6	1251.9	719.9	389.2	1.2	67.4	34.8	6.0
	Mean	4878.2	1462.2	534.7	384.7	1.7	60.7	40.5	9.2
	1	1744.2	153.4	544.3	52.6	0.7	11.9	24.9	6.6
apaq	2	1938.5	1044.4	589.0	346.7	1.2	39.7	29.9	11.5
Mankabad	3	1479.3	629.6	699.9	382.0	1.8	24.3	30.8	7.3
	Mean	1720.7	609.1	611.1	260.4	1.2	25.3	28.5	8.5
	1	1523.1	251.0	713.5	355.9	1.6	22.9	43.9	15.4
, and	2	2558.0	633.0	597.0	355.5	1.8	35.7	33.4	7.6
Ellwan	3	1634.0	388.5	719.6	300.6	1.0	17.8	39.1	15.8
	Mean	1905.0	424.2	676.7	337.3	1.5	25.5	38.8	12.9
	1	2008.0	120.9	598.7	207.9	0.8	17.1	47.8	13.9
ussie	2	1463.1	144.5	743.7	212.0	0.7	22.2	34.6	8.2
Bani Hussien	3	1899.9	92.7	543.3	164.9	0.5	20.0	33.2	13.9
l eg	Mean	1790.3	119.4	628.5	194.9	0.7	19.8	38.5	12.0
_	1	1184.4	223.2	488.9	134.1	0.8	16.1	32.9	18.0
halit	2	4596.7	133.6	457.5	134.6	0.8	12.0	34.5	9.2
Bani Ghalib	3	2841.8	156.9	456.7	136.5	1.3	10.6	32.3	18.1
	Mean	2874.3	171.2	467.7	135.1	1.0	12.9	33.2	15.1
	1	1180.5	111.8	440.9	104.2	0.5	13.1	29.2	15.5
Jam	2	1361.0	149.1	480.6	105.7	0.4	11.7	37.4	15.1
Gahdam	3	1007.3	164.3	464.0	156.6	0.5	16.8	36.0	8.3
	Mean	1183.0	141.7	461.8	122.2	0.5	13.9	34.2	13.0

Table 4: continue.

Vilage	Sites	Fe	Zn	Mn	Cu	Cd	Pb	Ni	Cr
					Subsurfa	ce Layer			
	1	3861.2	1085.7	462.7	396.7	1.2	40.5	45.1	41.5
ged	2	2019.2	574.7	394.1	150.8	0.8	13.7	34.5	18.4
Madabeg	3	2701.1	458.5	436.5	171.1	0.9	28.7	39.5	20.2
-	Mean	2860.5	706.3	431.1	239.5	0.9	27.6	39.7	26.7
	1	1318.6	118.6	457.7	140.5	1.1	9.0	30.1	17.8
abad	2	1818.6	218.4	589.1	293.7	0.9	15.7	32.4	19.6
Mankabad	3	1437.9	195.3	616.5	249.5	1.4	12.4	30.6	18.1
	Mean	1525.1	177.4	554.4	227.9	1.1	12.3	31.0	18.5
	l	1716.3	139.5	691.0	238.0	1.1	13.1	34.5	13.4
,an	2	2286.6	251.6	582.4	277.6	1.2	16.2	47.3	16.7
Ellwan	3	1549.0	177.3	554.7	273.3	0.9	9.6	39.1	24.5
	Mean	1850.6	189.5	609.4	263.0	1.1	13.0	40.3	18.2
	1	1902.0	173.4	608.3	205.4	1.0	22.7	44.4	16.6
Bani Hussien	2	1375.2	86.6	469.1	184.9	0.8	17.8	41.8	40.1
H im	3	2165.7	105.7	600.6	140.0	0.7	15.2	38.5	20.0
Ä	Mean	1814.3	121.9	559.3	176.8	0.8	18.6	41.6	25.6 .
	1	1141.1	148.7	591.1	115.7	0.7	13.4	42.3	13.5
Bani Ghalib	2	1139.9	102.8	480.9	95.2	0.6	9.6	34.5	29.8
ani O	3	1252.1	137.7	428.4	115.3	0.9	17.0	24.8	14.8
<u> </u>	Mean	1177.7	129.7	500.1	108.7	0.7	13.3	33.9	19.4
	1	1407.1	113.9	471.4	97.1	0.7	16.1	40.1	37.4
lam	2	1624.4	112.9	461.1	86.6	0.5	13.2	29.2	15.5
Gahdam	3	1150.5	143.6	514.9	156.6	0.7	14.6	41.0	13.6
	Mean	1394.0	123.4	482.5	113.4	0.6	14.6	36.8	22.2

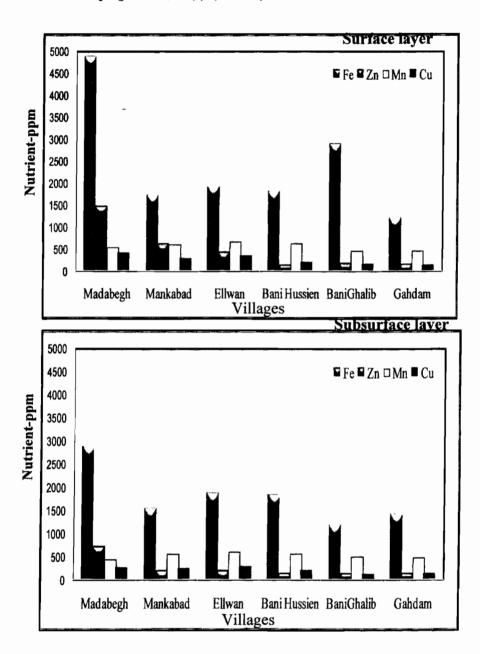
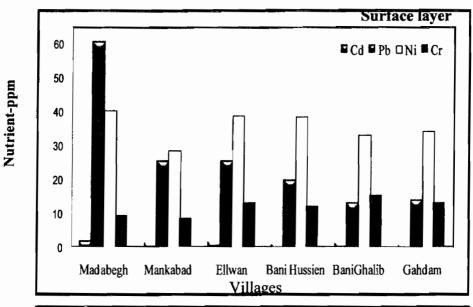


Fig.(1): Variation in total contents of micronutrients - ppm (Fe, Zn, Mn, and Cu) in the soils of the studied six locations irrigated with sewage waste water.

Nutrient-ppm



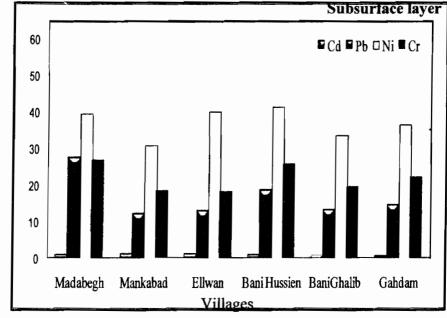


Fig.(2): Variation in total contents of micronutrients - ppm (Cd, Pb, Ni, and Cr) in the different layers of the studied six locations irrigated with sewage waste water.

Interrelationships among total element contents and soil properties:

The relationships between the total concentrations of metals in different locations of the polluted area and soil properties are summarized in Table (5). Generally, significant negative correlation coefficients were obtained between total of each Fe, Zn, Cu, Cd, Pb and Cr in soil and soil pH. The correlation

coefficients between soil pH and total amount of Fe, Zn, Cu, Cd, Pb and Cr were -0.681", -0.675**, -0.525*, -0.560*, -0.644* and -0.632**, respectively. The CEC was significantly positive correlated with total amounts of Fe and Mn (The r values were 0.661**, 0.626 and respectively), and negatively correlated with total Zn (r= -0.514**).

Table(5):Correlation coefficients between total amounts of micronutrients and heavy metals, and soil properties of the soils of the studied area.

Total	Depth	EC	pН	Clay	Silt	Sand	CEC	O.M.
Fe	0-25	-0.101	-0.467	-0.312	-0.242	0.341	-0.626**	0.017
	25-50	-0.412	-0.681**	-0.336	-0.462	0.467	-0.339	-0.207
Zn	0-25	-0.243	-0.675**	-0.068	-0.339	0.313	-0.447	0.038
211	25-50	-0.346	-0.601**	-0.502*	-0.521°	0.573*	-0.514°	-0.372
Mn	0-25	-0.440	0.146	0.320	0.303	-0.396	0.347	0.138
17411	25-50	0.035	0.247	0.389	0.493*	-0.510*	0.661**	0.187
Cu	0-25	-0.378	-0.525*	0.060	0.014	-0.039	-0.005	-0.053
Cu	25-50	-0.430	-0.554°	-0.087	0.049	-0.003	0.039	-0.320
Cd	0-25	-0.343	-0.560°	-0.073	-0.065	0.087	-0.150	-0.032
Cu	25-50	-0.234	-0.174	0.052	0.170	-0.145	-0.002	-0.202
Pb	0-25	-0.212	-0.644**	-0.107	-0.278	0.280	-0.346	-0.102
10	25-50	-0.149	-0.661**	-0.294	-0.302	0.333	-0.215	-0.240
Ni	0-25	-0.334	-0.445	-0.073	0.017	0.019	0.035	0.267
	25-50	-0.232	-0.365	-0.024	0.093	-0.059	0.426	0.334
Cr	0-25	0.058	-0.019	-0.277	0.034	0.097	0.327	-0.048
	25-50	-0.475*	-0.632**	-0.008	-0.197	0.148	-0.199	0.017

^{*,** =} p < 0.05 and 0.01, respectively.

Extractability of different extraction procedures:

The objective of this part is to explore the extractability of some extraction procedures that widely used in determining the amount of mobile forms of heavy metals in polluted soils irrigated with untreated sewage wastewaters.

Comparing the extraction procedures for their ability to extract the potential of each heavy metal to enter the mobile pools and becoming environmentally hazard is undertaken.

The amounts of extractable Fe. Mn, Zn, Cu, Cd, Ni, Pb and Cr in the 18 soils via the tested extractants varied depending on extractants and used. metals Among the extractants. EDTA-AAAc the extracted proportion of Fe, Zn, Mn, Cu, Cd and Pb followed by EDTA-Ammonium Citrate with Fe, Zn, Mn, Cu and Pb, while HCl + H₂SO₄ were the lowest for the entire eight studied elements. Regardless of soils and extractants, relative extractability was higher for Ni and Cr as compared to other two heavy metals (Table 6 and Figure 3).

Table (7) showed the correlation coefficients for soil рH extracted Zn. and Pb. These relations showed that extracted Zn. with HCl+H₂SO₄, EDTA, EDTA-EDTA-Ammonium AAAc and Carbonate extractants correlated highly significantly and negatively with soil pH. At the same time the extracted Zn by HC1+H₂SO₄, EDTA and EDTA- AAAc were

significantly correlated and CEC. negatively with soil However, extracted Pb with AB-DTPA and EDTA-AAc correlated significantly and negatively with soil CEC; the r values were -0.52* -0.489*, receptively. relations presented in Table (7) showed that the concentrations of DTPA, and HCl + H₂SO₄ Mn were negatively significantly and correlated with soil pH. Also, Mn extracted with AB-DTPA **EDTA** and were positively significantly with soil CEC.

In the present study, extracted Cu with DTPA, EDTA, EDTA-EDTA-AAAc, AAc. EDTA-Ammonium Carbonate and EDTA-Ammonium Citrate at each site of the studied area were strongly related to soil pH (r = -0.511*, -0.479*, -0.492*, -0.543*, -0.654** and -0.556*, respectively). It was found that soil pH of studied area highly significant and negatively correlated with EDTA-AAAc and HCl+H2SO4 extracted Cd and Ni, Table (7). Cation exchangeable capacity of the tested soils was correlated significantly and positively with EDTA-AAc -Cd, while CEC was correlated significantly and negatively with EDTA Cd. Correlation coefficients between CEC of the polluted soils and extracted Ni with EDTA-AAAc, EDTA. EDTA-Ammonium Citrate and HCl+H₂SO₄ are presented in Table (7). The data of these extractants showed a good and significant correlation with soil CEC.

Table(6): Minimum, maximum, and averages of extracted amounts of micronutrients and heavy metals (ppm) using different extraction procedures.

		Τ		Т	T		T .	T	T	Τ
Extractant		Depth	Fe	Zn	Mn	Cu	Cd	Pb	Ni	Cr
	Min	0-25	0.00	0.00	0.00	0.20	0.00	0.00	0.00	0.00
	Min ₂	-50	0.00	0.04	0.04	0.08	0.00	0.00	0.00	0.00
HCl+ H₂SO₄	Max	0-25	2.44	0.80	9.86	0.96	0.26	1.23	1.41	0.40
		25-50	0.52	1.96	8.60	0.68	0.07	0.97	0.77	0.20
DTPA- Amm.Bicarbonate EDTA	Aver.	0-25	0.24	0.24	4.72	0.38	0.04	0.21	0.22	0.02
	711011	25-50	0.09	0.53	3.70	0.31	0.03	0.18	0.12	0.02
	Min ₂ :	0-25	0.45	0.97	2.16	0.15	0.00	0.00	0.33	0.00
	2:	-50	0.42	0.90	1.16	0.12	0.00	0.00	0.49	0.00
DTPA	Max	0-25	5.95	9.92	7.58	3.59	1.13	3.00	0.74	0.71
		25-50	2.78	6.39	6.40	2.25	0.11	1.96	0.91	0.69
	Aver.	0-25	2.87	2.78	4.60	1.62	0.36	0.93	0.59	0.26
	ļ	25-50	1.03	1.94	2.98	1.00	0.03	0.58	0.65	0.30
	Min ₂₅	0-25	1.75	0.65	0.94	0.29	0.00	0.19	0.00	0.00
	25	-50	1.92	0.61	0.00	0.47	0.00	0.19	0.00	0.03
	Max	0-25	19.11	3.35	7.59	3.59	0.03	5.70	0.21	0.20
Amm.Bicarbonate		25-50	9.38	8.58	8.48	4.02	0.04	3.13	0.22	0.17
	Aver.	0-25	5.06	1.34	3.83	2.12	0.01	1.67	0.09	0.09
	711011	25-50	3.92	1.94	4.48	2.46	0.01	0.89	0.09	0.11
	Min ₂₅	0-25	0.00	1.98	1.43	1.50	0.06	2.86	0.45	0.00
	25		0.00	0.00	1.39	1.94	0.01	2.02	0.90	0.00
EDTA	Max	0-25	5.86	7.97	25.00	10.05	0.97	9.92	3.31	0.44
55111		25-50	6.42	9.92	17.57	13.56	1.09	17.66	2.80	1.69
	Aver.	0-25	4.31	4.20	10.29	6.95	0.54	5.85	1.99	0.02
		25-50	3.25	3.90	8.96	6.63	0.38	7.34	2.13	0.09
	Min ₂₅	0-25	1.68	1.50	0.00	0.08	0.21	0.37	0.39	0.00
	25	-50	4.41	0.87	0.69	1.00	0.00	0.02	0.16	0.00
EDTA- AAc	Max	0-25	9.21	14.24	7.70	24.67	0.95_	29.19	1.58	1.85
2011114		25-50	20.90	12.71	6.50	17.52	0.80	8.19	2.24	3.32
	Aver.	0-25	5.34	6.34	3.26	8.55	0.55	6.81	0.91	0.31
	11101.	25-50	8.91	4.68	3.04	6.55	0.41	3.75	0.97	0.69
	Min ₂₅	0-25	2.69	1.25	11.25	0.80	0.00	0.46	0.17	0.30
	25	-50	1.28	1.22	11.50	0.49	0.00	0.40	0.22_	0.42
EDTA- AAAc	Max	0-25	25.94	101.15	49.60	15.47	0.32	27.38	1.35	0.78
		25-50	14.50	33.80	42.10	8.59	0.15	8.94	1.12	2.98
	Aver.	0-25	10.16	20.37	27.64	7.18	0.06	7.47	0.44	0.41
		25-50	8.38	5.71	28.27	4.75	0.04	3.71	0.44	0.98
	Min ₂₅	0-25	2.93	0.80	6.57	1.21	0.00	0.00	0.00	0.00
	25	-50	2.99	0.43	7.24	0.65	0.00	0.00	0.00	0.00
EDTA-	Max	0-25	85.24	19.49	24.88	11.96	0.90	8.90	1.13	0.94
Amm. Carbonate	IVIAX	25-50	34.41	5.41	14.20	9.49	0.13	3.14	0.96	1.66
	Aver	0-25	20.03	5.50	12.54	4.77	0.11	2.18	0.25	0.39
	Aver.	25-50	8.61	2.02	10.45	3.55	0.03	1.53	0.31	0.42
	Min	0-25	5.03	1.93	13.35	1.03	0.03	1.07	0.12	0.35
	Min ₂₅	-50	6.97	1.18	13.45	0.22	0.03	1.04	0.15	0.35
EDTA-	Max	0-25	42.94	56.80	47.15	18.86	0.59	23.82	0.38	0.42
Amm. Citrate		25-50	23.07	55.60	41.45	9.98	0.15	15.73	0.60	2.00
	(0-25	16.58	14.83	33.40	5.96	0.31	7.50	0.20	0.38
	Aver.	25-50	13.03	6.88	31.07	4.82	0.07	4.11	0.26	0.87
		23-30	13.03	0.00	31.07	4.02	0.07	4.11	0.20	0.07

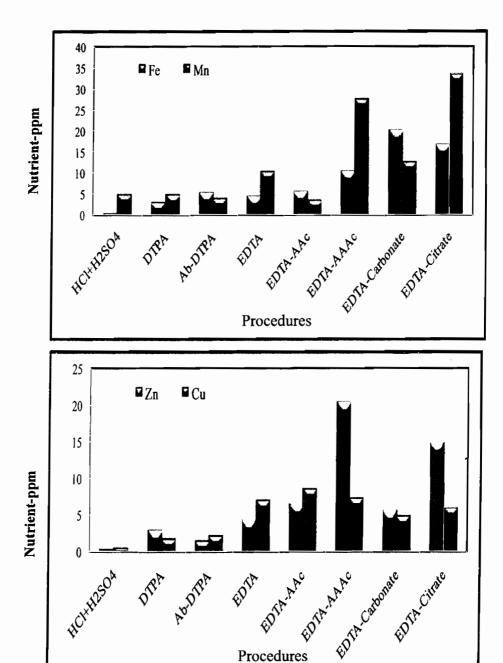


Fig. (3): The concentrations of extractable heavy metals (ppm) in 18 soils by the eight tested extractions.

Procedures

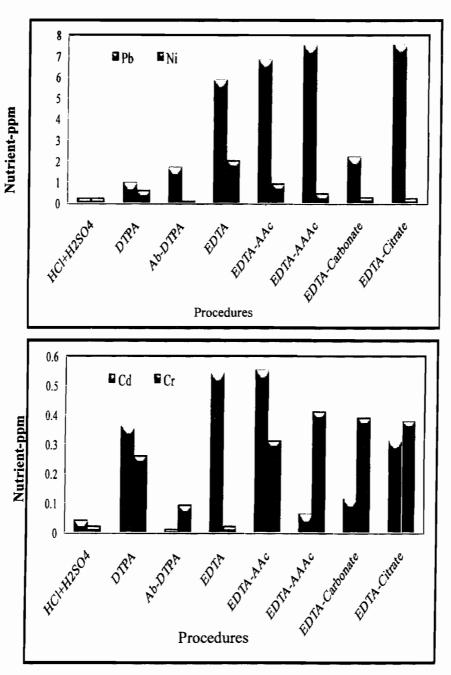


Fig. (3): Continue

Table(7): Correlation Coefficients between extracted heavy metals in the studied area and soil (pH and CEC) in the surface soils

layer:

	layer.							
pН	Fe	Zn	Mn	Cu	Cd	Pb	Ni	
				Soil pH	,			
HCl+H ₂ SO ₄	0.359	0.418	-0.631**	-0.556*	-0.628**	0.120	-0.632**	0.014
DTPA	-0.563*	-0.752**	-0.621**	-0.511°	-0.464	-0.699**	-0.030	-0.240
Ab-DTPA	-0.538°	-0.719**	0.340	-0.284	-0.585*	-0.580*	-0.379	0.247
EDTA	-0.496*	-0.187	-0.108	-0.041	-0.053	0.307	-0.015	-0.007
EDTA-Aac	0.443	-0.276	-0.025	-0.492*	-0.090	-0.515*	0.309	0.234
EDTA- AAAc	-0.718**	-0.708**	0.103	-0.479*	-0.535°	-0.653**	-0.659**	-0.036
EDTA-Amm. Carbonate	-0.124	-0.664**	-0.173	-0.543*	-0.082	-0.710**	0.048	-0.359
EDTA-Amm. Citrate	-0.742**	-0.600**	0.678**	-0.654**	-0.295	-0.536*	-0.068	-0.330
			Sc	oil CEC				
HCI+H ₂ SO ₄	-0.218	0.287	-0.008	-0.120	-0.353	0.191	-0.564*	0.283
DTPA	0.117	-0.446	-0.356	-0.342	-0.048	-0.398	0.247	0.238
Ab-DTPA	-0.452	-0.504*	0.641**	0.028	-0.443	-0.402	-0.381	-0.317
EDTA	-0.044	0.049	0.599**	0.307	-0.101	0.380	0.638**	0.263
EDTA-Aac	-0.006	-0.013	0.408	-0.075	-0.242	-0.381	0.236	0.212
EDTA- AAAc	-0.316	-0.471*	-0.102	-0.080	0.027	-0.385	-0.560*	-0,241
EDTA-Amm. Carbonate	-0.659**	-0.518*	-0.004	-0.221	-0.158	-0.327	-0.265	0.133
EDTA-Amm. Citrate	-0.268	-0.420	0.316	-0.132	-0.147	-0.429	-0.552*	-0.416

^{*, **=} significant at: p < 0.05, p < 0.01 and p < 0.001 probability levels. All other p > 0.05.

Generally, and depending on the previous results, it could be stated that the soil pH and CEC of the soils are the main soil factors which affected on most of the heavy metal contents of the soil which extracted with various extractants.

Part 2

This part of the research work devoted to study the suitability of

extraction procedures for predicting the levels of available micronutrients and heavy metals in polluted soils.

In all soil after planting the extractable Zn, Cu and Pb from soil samples were decreased, but the extractable amounts of other metals (Fe, Mn, Cd, Ni and Cr) were remarkably increased after corn harvesting. However, after wheat harvesting, in all soil locations, the amount of all extractable metals, but not Zn, Cd, Ni and Cr, were decreased comparing either to the initial amounts before planting or to the amounts after corn harvesting. The mean values of extractable Fe. Zn. Mn. Cu. Cd. and Pb were decreased. It should be noticed that the changes in extractable amounts of metals after corn harvesting are little higher than that after wheat harvesting. These changes reflect the effect of plant the status oftype on micronutrients and heavy metals in root zone. The changes in the amounts of extractable metals could be attributed either to the variations in crop requirements micronutrients and heavy metals, or to the variations in the ability of corn and wheat root system and root exudates to the metals from the release exchange sites, or dissolve the metals from their bearing compounds. Root system and root exudates both exert great influences on solubility and release of nutrients into root

zone, thus increase the amounts extractable metals. thought both corn and wheat crops are C4 plants, their root systems are markedly differed. and exert different effects on the of status metals in surrounding root zone. Therefore, good care should be taken, when deciding which element will be determinate. Soil type and the type of crop plants should be considered when deciding which extraction procedure should use.

Data summarized in Table 8 show the amounts of extractable micronutrients and heavy metals by eight different extracted extraction procedures from polluted soils before planting, and at the end of growth seasons of corn and wheat. The ability of the extraction procedures were varied with the type of extractant, the type of polluted metals, and the type of previous planted crop. For instance, with respecting to the extractable amount of Fe. before planting EDTA-AAAc gave the highest amount, while after corn and wheat EDTA gave the highest amount. In general, the amounts of metals extracted with HCL+H2SO4 were the lowest, either before planting or after growing of corn and wheat crop. The EDTA extraction procedures, on the other hand, were most efficient the procedures for extracting the micronutrients and heavy metals from the soils irrigated with sewage wastewater.

Table(8): The effects of extraction procedures on the amount of

extracted metals (ppm) during the growth seasons.

Extraction Procedures		Zn	Mn	Cu	Cd	Pb	Ni	Cr		
			Before P					,		
HCl+H2SO4	0.24	0.24	4.72	0.38	0.04	0.21	0.22	0.02		
DTPA	2.87	2.78	4.60	1.62	0.36	0.93	0.59	0.26		
Ab-DTPA	5.06	1.34	3.83	2.12	0.01	1.67	0.09	0.09		
EDTA	4.31	4.20	10.29	6.95	0.54	5.85	1.99	0.02		
EDTA-Aac	5.34	6.34	3.26	8.55	0.55	6.81	0.91	0.31		
EDTA-AAAc	10.16	20.37	27.64	7.18	0.06	7.47	0.44	0.41		
EDTA-Carbonate	20.03	5.50	12.54	4.77	0.11	2.18	0.25	0.39		
EDTA-Amm.Citrate	16.58	14.83	33.40	5.96	0.31	7.50	0.20	0.38		
Mean	8.07	6.95	12.53	4.69	0.25	4.08	0.59	0.24		
LSD _{0.05}	4.66**	4.23**	3.61**	1.86**	0.16**	1.99**	0.16*	0.17**		
After Corn										
HCL+H2SO4	0.11	0.26	7.38	0.36	2.25	1.46	0.18	0.00		
DTPA	16.31	5.40	15.13	11.25	4.71	5.02	2.32	0.00		
Ab-DTPA	32.21	2.15	6.81	4.65	1.53	2.73	1.20	0.35		
EDTA	54.63	4.40	13.34	2.83	4.11	4.29	1.85	1.39		
EDTA-AAc	22.13	8.92	4.73	1.11	7.41	2.19	1.14	4.36		
EDTA-AAAc	13.78	3.00	13.66	1.64	1.66	3.25	2.24	0.00		
EDTA-Carbobate	2.91	3.82	19.17	1.17	1.23	1.82	0.41	0.12		
EDTA-Amm.Citrate	8.04	3.22	39.16	1.20	1.36	1.53	3.06	1.86		
Mean	18.77	3.90	14.92	3.03	3.07	2.79	1.55	1.01		
LSD _{0.05}	6.30**	1.12**	2.66**	0.79**	0.52**	0.69**	1.25**	2.16**		
			After W	heat						
HCL+H2SO4	1.12	0.21	0.96	0.08	0.38	0.07	0.38	0.24		
DTPA	4.59	2.04	4.87	3.42	2.79	1.94	0.20	0.58		
Ab-DTPA	11.9	2.64	8.18	4.04	3.91	6.64	0.16	0.48		
EDT'A	19.91	5.47	18.55	2.85	4.97	5.00	4.16	0.74		
EDTA-AAc	8.45	12.88	20.30	0.63	1.03	0.95	0.54	0.27		
EDTA-AAAc	9.06	4.38	10.42	1.37	1.57	2.68	1.80	1.32		
EDTA-Carbobate	2.39	14.52	20.02	1.18	1.33	1.79	1.45	0.79		
EDTA-Amm.Citrate	5.72	5.02	10.40	4.23	9.55	1.92	0.50	0.00		
Mean	7.54	5.89	11.71	2.22	3.19	2.62	1.15	0.55		
LSD _{0,05}	2.49**	4.93**	2.64**	0.56**	0.97**	0.60**	1.09**	0.46**		

^{*} and ** significant at probability level <0.05 and 0.01, respectively n.s not significant

also greatly varied in predicting with changing the type of crop the changes in the status of plants. The EDTA- procedures

Extraction procedures were metals in root zone associated

show the highest variations in the extractable amounts of most metals, especially Fe, Mn, Cd and Ni. For instance, the amounts of extracted Fe by EDTA were changed from 4.31 ppm before planting to 54.63 and 19.91 ppm after corn and wheat harvesting. respectively. However, in the case of Zn the extracted amounts were changing from 4.20 ppm before planting to only 4.40 and 5.47 ppm after corn and wheat, respectively. Using acetate ions or acetic acids with EDTA improve its ability to predict the variations in the amounts of available metals.

Nyamangara, and Mzezewa (1999) reported that extraction procedure is used to extract the bioavailable fractions metal in sewage contaminated area. The release of electrostatically metal ions weakly bound exchange to surfaces is promoted by ion exchange with cations such as ammonium (NH₄⁺) (Ure, 1996). Moreover, including the acetate ions and/or acetic acids and ammonium ions with enhances its ability to extract labile (exchangeable and soluble) form of metals, and improves its ability to predict the variations in the amounts of available metals. This is quite obvious in the case of Fe (Table 8) .EDTA-Ammonium forms strong complexes with numerous heavy metals and release their exchangeable forms (KabataPendias and Pendias, 1984), and organically complexes "pools" (Podlesakova *et al.*, 2001).

The presence of acetate ions with EDTA also promotes the ability of EDTA to extract metals from polluted soils. The data presented in Table 8 show that in all studied growth seasons the highest values of all extractable metals, but Cu, Cd, and Pb, were obtained by EDTA-AAAc (pH 4.65), while the lowest were found with HCL+H₂SO₄. superiority of EDTA-AAAc as compared to other extractants may be duo to its low pH (4.65) and the chelating effects of acetate anions, each of the two characters may enhanced metals extractability (Hegazy, 1980; Hegazy et al., 1991; and Abd El-Haleem et al., 2002).

DTPA The extraction procedures, on the other hand, extract low amounts of heavy micronutrients and compared to EDTA extraction procedures (Table 8). instance, before planting, amounts of Fe, Zn, Mn, and Cu extracted by DTPA were 2.87, 2.78. 4.6. and 1.62 respectively, compared to 4.31, 10.20, and 6.95 extracted by EDTA, respectively. The same thing is quite true for heavy metals, especially Cd, Cr and Pb. This may be due to what mentioned have been Bermond et al. (1998) that the EDTA procedures are specific extraction, and they can

extract labile and non-labile fractions of micronutrients and heavy metals. Nomeda et al. (2004) reported that extractant's pH has a considerable influence on extractable heavy metals.

Inspecting our results emphasized that choosing the extraction procedures studying the status of heavy metals pollutants in soils irrigated for a long time with wastewater requires taking into account the aim of the investigation, the soil type, the source of the contaminants, as well as the agricultural practices and the nature of the previous crops. These findings are in agreement with those reported by Brazauskine et al. (2004), and Gupta and Sinha (2006).

Metals concentrations in corn and wheat plants grown on soils irrigated with sewage wastewater for long time:

Concentrations of heavy metals and micronutrients in corn and wheat plants grown on soils irrigated with sewage wastewater for long time are considerably varied with the variations in the amounts of available metals studied soils in among the different locations. concentrations of Fe, Pb, Ni, and Cr in corn plants were decreased with increasing the distance from the source-point of pollution (Madabegh). For instance, the concentrations of Ni in corn on soils plants grown (Madabegh and Mankabad) are

noticeably higher than those of Bani-Ghalib and Gahdam. Also corn plants grown on soils of Bani-Ghalib, which is far away source the point pollution, contained the lowest amount of Cr (3.40 ppm) as compared with those grown on soils of Madabegh, Ellwan, Mankabad. Bani Hussein, and Gahdam (9.42,23.71, 20.64, 16.78, and 10.23 respectively). variations in the concentrations of the other metals among locations did not show obvious trends (Table 9). The concentrations of all studied heavy metals in corn plants, but not Pb in some locations, are higher than the (accepted ranges). The concentrations of Pb in corn plants grown on the soils of Ellwan. Bani Hussein. Gahdam were lower than the normal level of 4 mg kg⁻¹ dry. matter as reported by Kabata Pendias and Pendias (1992).

degree, To some the variations in metals concentrations in corn plants are in agreement with the variations in the available amounts of metals in soils. The disagreements, that might be noticed, between the concentrations of heavy metals in corn plants and the amounts of extracted metals from soils of the studied location could be due to the variability in the potentials of the different extraction procedures used in this study.

Table(9): Dry matter accumulation (gm/pot⁻¹) and concentrations of micronutrients and heavy metals in corn plants grown on polluted soils of the studied locations[®].

Locations (Villages)	DM gm	Nutrient - ppm							
(Villages)	pot 1	Fe	Zn	Mn	Cu	Cd	Pb	Ni	Cr
Madabegh	4.34	879.07	34.82	54.79	18.29	7.84	21.53	18.81	9.42
Mankabad	6.53	800.49	28.30	55.53	16.67	12.43	4.70	17.93	23.71
Ellwan	4.58	470.23	36.66	36.66	20.19	30.13	1.54	14.86	20.64
Bani Hussien	6.26	643.51	29.42	55.52	19.88	6.93	1.68	12.33	16.78
Bani Ghalib	4.40	585.59	22.82	64.02	18.64	8.92	10.51	2.53	3.40
Gahdam	3.54	609.99	25.49	57.77	20.11	16.90	0.99	4.20	10.23
Mean		664.81	29.58	54.05	18.96	13.86	8.49	10.11	14.03
LSD _{0.05}		110.76**	4.50**	8.21**	n.s.	15.76*	5.36**	4.70**	6.02**

^{*} and ** significant at probability level ≤ 0.05 and 0.01, respectively n.s not significant [@] Each value represents a mean of three sites

The concentrations of Fe, Mn, Cu, Cd, and Pb, in wheat straw. significant differences between locations (Table 10). However, the variations concentrations of all studied metals, but not Cu, in wheat seeds were significant. Straw and seeds of wheat plants grown on locations near to the source point sewage-waste of (Madabegh and Mankabad) show concentrations high of micronutrients and heavy metals compared to those far away from the source point of pollution. Lead concentrations. example, in straw of wheat plants grown on soils of Bani-Ghalib and Gahdam were 8.67 and 7.03 ppm, respectively, compared to 15.6 and 13.8 ppm in straw of wheat plants grown on the soils

of Madabegh and Mankabad, respectively. In general view, the concentrations of heavy metals and micronutrients in wheat plants either significantly decreased or show no significant variations with increasing the distance from the source point of pollutants. This is, in general, in agreements with the variations in extractable heavy metals and micronutrients between soils of different locations. concentrations of heavy metals in most plants grown on the studied locations were decreased with increasing the distance from Madabegh and Mankabad. As mentioned before, this is a direct result of decreasing the amounts of heavy metals pollutant in the down-stream of the irrigation cannel.

Table(10): Dry matter accumulation (gm/pot⁻¹) and concentrations of micronutrients and heavy metals in straw and seeds of wheat plants grown on polluted soils of the studied locations.

	DM				Nutrien	t - ppm						
Locations (Villages)	gm pot	Fe	Zn	Mn	Cu	Cd	Pb	Ni	Cr			
	Wheat straw											
Madabegh	3.38	730.81	43.57	56.04	16.34	13.80	15.60	14.90	20.12			
Mankabad	3.66	720.92	34.52	54.44	21.22	6.64	13.80	10.08	18.86			
Ellwan	3.09	586.62	30.48	49.53	27.39	9.75	8.40	14.06	19.34			
Bani Hussien	3.71	666.80	30.55	43.92	25.39	11.23	7.13	10.41	20.59			
Bani Ghalib	3.58	662.68	29.17	42.35	28.33	12.19	8.67	12.84	18.39			
Gahdam	3.58	511.79	33.54	75.88	23.38	15.19	7.03	11.76	20.29			
Mean	3.50	646.60	33.64	53.69	23.68	11.47	10.10	12.34	19.60			
LSD _{0.05}		115.69**	4.84**	n.s.	3.78**	3.44**	4.96**	n.s.	n.s.			
				Wheat	seed							
Madabegh	3.39	287.10	28.08	24.65	12.52	7.27	7.27	2.52	5.44			
Mankabad	3.29	435.46	34.35	24.66	13.57	5.58	5.58	8.95	8.05			
Ellwan	2.70	325.18	36.98	19.64	13.32	4.96	4.96	1.60	10.62			
Bani Hussien	3.81	377.96	30.54	26.55	14.19	7.76	7.76	11.83	2.67			
Bani Ghalib	3.34	295.47	35.60	35.53	11.84	6.83	6.83	2.51	2.23			
Gahdam	3.60	317.17	40.38	29.44	11.26	6.13	6.13	2.83	7.84			
Mean	3.35	339.72	34.32	26.74	12.78	6.42	6.42	5.04	6.14			
LSD _{0.05}		90.69*	4.67**	5.78**	n.s.	1.51**	1.51**	3.60**	4.00**			

* and ** significant at probability level ≤ 0.05 and 0.01, respectively n.s not significant. [@] Each value represents a mean of three sites

The concentrations of metals in reproductive organs, where plants tend to store the metabolic products, may be more sensitive the variations in concentrations of metals in root zone. Therefore, regarding to study the effects of heavy metal pollutants on plants, concentrations of these pollutants reproductive plant parts, especially seeds should considered well. For instance, as

presented in Table 10, the concentrations of Cd in seeds of wheat plants grown on the soils of all studied locations ranged between 4.96 and 7.76 with an average of 6.42 ppm, which is higher than the normal range of Cd concentrations (0.9 – 2.6 mg kg⁻¹ DM) as reported by Kabata-Pendias and Pendias (1992). In general, it may be stated that concentrations of Zn and Mn in seeds of wheat plants grown on

the soils of the studied region were in the normal range. Contrary to that. the concentrations of Fe, Cu, Cd, Pb, Ni. and Cr in seeds of wheat plants grown on these soils were higher than the normal ranges reported by Kabata Pendias and Pendias (1992).

Judging the extraction procedures suitable for micronutrients and heavy metals:

Melsted and Peck. (1973) defined the available form of nutrient by saying "By available nutrient one usually means the chemical form or forms of an essential plant nutrient in soil whose variation in amount is reflecting in variations in plant growth and plant yield". In polluted soils there are many forms of metals with different degree of solubility's that may be enable them to enter the available pool at any time. Therefore, the extraction procedures used for studying the presence of metals pollutants in soils at any time should have a good association with one of the growth responses whose variations are associated with the variations in the size of available pool. Nutrient the uptake by plants is considered here as one of those crop responses that could be used to judge the suitability of the extraction procedure. The studied procedures for extracting the available amounts micronutrients and heavy metals

from polluted soils were judged using the correlations coefficients between the amounts of extracted metals and metal uptake by corn and wheat plants grown on these soils. Table 11 and 12 represent the correlation coefficients between extractable metals by each of the studied extraction procedures and the metal uptake in corn and wheat plants grown on the soils of the studied area.

of all. HCI+H2SO4 extraction procedure did not gave any significant correlation with amounts of micronutrients and heavy metals, but Cd, taken up by either corn and wheat plants, the results which indicated that this procedures is not a suitable one for extracting metals from polluted soils. This extraction gave procedure the lowest amounts of micronutrients and heavy metals, and did not predict the availability of the studied well. Therefore. metals the correlation coefficients between the extracted metals and metals uptake by corn and wheat plants were small and not significant (at p levels ≤ 0.05). Moreover, the results of this procedure were not persistent and show wide variations between soils and after the harvesting of different crops, therefore the HCl+H2SO4 extraction procedure could not be used successfully for predicting the status of metal pollutants in soils. Cox and Kamprath (1972) stated that one of the criteria which the extraction upon

procedures is selected is that "the extracted amount should be correlated with the growth and response of each crop to that nutrient under various conditions".

DTPA extraction procedure gave significant correlations with Fe, Mn, Zn, Cu, and Pb uptake in corn stover (Table 11). In the wheat plants case procedure did not give significant correlations with any of the studied metals taken up in however. wheat straw, significant procedure gave correlation with Fe, Mn, Zn, Cu, and Cd in wheat seeds, and with total uptake of Fe, Zn, and Cu (Table 12). On the other hand, the DTPA extraction procedure did not give any significant correlations with the amounts of any one of studied heavy metals either in corn or in wheat plants.

The DTPA micronutrient extraction method, developed by Lindsay and Norvell (1978), is a

nonequilibrium extraction estimating the potential bioavailability of Zn, Cu, Mn, and Fe for neutral and calcareous soils. It is, in general, considered many researchers specialized procedure. Vose and Randall (1962) stated that the three important metals pools in micronutrients supplying plant during the season are (i) H₂O-soluble; (ii) exchangeable; and (iii) adsorbed, chelated or complexed. extraction procedure for micronutrients, therefore, should extract a portion or all of these pools. The well the procedure predicts overall effects of these three pools on the availability of Fe, Mn, Zn, and Cu in normal soils. Therefore, using the DTPA to predict the amounts of other heavy metals in polluted soils may not be correct, the obtained. results revealed that.

Table(11): Correlation coefficients between extractable metals by different extraction procedures and metals uptake by corn

	piants.							
Extractants	Fe	Zn	Mn	Cu	Cd	Pb	Ni	Cr
	1		Corn s	tover				
HCl+H₂SO₄	-0.23	-0.15	-0.09	-0.09	-0.02	-0.10	-0.09	
DTPA	0.44**	0.72**	0.43**	0.27*	0.11	0.26*	0.18	
Ab-DTPA	0.47**	0.63**	-0.25*	0.03	-0.20	-0.22	0.07	-0.24*
EDTA	0.02	0.34**	0.10	0.56**	0.62**	0.26*	0.17	-0.16
EDTA-AAc	-0.08	0.21	-0.09	0.11	0.30**	-0.04	0.29*	0.26*
EDTA-AAAc	0.32**	0.44**	0.30**	0.32**	0.18	0.34**	0.06	
EDTA-Carbonate	0	0.04	-0.02	0.10	0.04	0.16	0.03	-0.10
EDTA-Citrate	-0.33**	0.28*	-0.30*	0.13	-0.07	0.19	-0.03	0.07

^{*, ** =} significant at: $p \le 0.05$ and 0.01 probability levels, respectively.

The performance of EDTA extraction procedures in predicting the bioavailability levels of metals in polluted soil

showed wide variations with the type of extracted metals, the type of plants, and the components of the extraction solutions.

Table(12): Correlation Coefficients between extractable heavy metals by different extraction procedures after wheat planting and wheat straw, seeds and total plant uptake:

data Wilett Biraw, Beeds and total prant apiane.								
Extractants	Fe	Zn	Mn	Cu	Cd	Pb	Ni	Cr
Wheat straw								
HCl+H ₂ SO ₄	-0.03	-0.19	0.03	0.00	0.242*	-0.06	0.12	0.09
DTPA	0.11	0.06	0.04	0.02	-0.05	0.02	0.06	-0.16
Ab-DTPA	-0.05	-0.16	-0.02	-0.13	0.15	0.22	0.13	-0.03
EDTA	-0.03	0.24*	-0.05	-0.33**	0.13	0.23	-0.13	0.20
EDTA-AAc	0.18	0.49**	0.04	-0.14	-0.01	0.34**	-0.21	-0.04
EDTA-AAAc	0.09	0.08	0.244*	-0.26*	0.02	0.26*	-0.04	-0.11
EDTA-Carbonate	0.14	0.55**	-0.17	-0.23*	0.08	0.31**	-0.04	-0.01
EDTA-Citrate	0.05	-0.08	0.25*	-0.23	0.12	-0.23	-0.12	
Wheat seed								
HCI+H₂SO₄	-0.09	0.09	0.11	0.02	0.28*	0.19	-0.05	0.11
DTPA	0.49**	0.63**	0.54**	0.53**	0.43**	0.12	-0.16	0.05
Ab-DTPA	0.34**	0.42**	-0.13	-0.06	-0.15	-0.03	0.01	0.15
EDTA	-0.17	-0.21	0.06	0.18	-0.14	0.11	0.28*	0.13
EDTA-AAc	-0.07	-0.21	-0.05	0.07	0.11	-0.13	0.10	-0.06
EDTA-AAAc	0.10	-0.12	-0.04	0.08	-0.01	0.03	0.30**	0.34**
EDTA-Carbonate	0.06	-0.15	-0.18	-0.05	-0.01	0.14	-0.04	0.04
EDTA-Citrate	-0.09	-0.07	-0.04	-0.14	0.05	-0 04	-0.13	
Whole wheat plant								
HCl+H₂SO₄	-0.07	-0.09	0.06	0.01	0.33**	0.08	0.04	0.14
DTPA	0.35**	0.41**	0.19	0.24*	0.16	0.09	-0.09	-0.09
Ab-DTPA	0.15	0.13	-0.05	-0.13	0.06	0.15	0.09	0.08
EDTA	-0.11	0.05	-0.03	-0.19	0.04	0.24*	0.14	0.24*
EDTA-AAc	0.09	0.22	0.02	-0.09	0.05	0.17	-0.06	-0.07
EDTA-AAAc	0.12	-0.01	0.22	-0.18	0.01	0.21	0.22	0.14
EDTA-Carbonate	0.13	0.30**	-0.21	-0.21	0.06	0.32**	-0.05	0.02
EDTA-Citrate	-0.01	-0.10	0.22	-0.25*	0.12	-0.20	-0.18	

^{*, ** =} significant at: p / 0.05 and 0.01 probability levels, respectively.

When the extraction solution contained EDTA only the extracted amounts of Mn, Cu, Cd, and Pb were significantly (p ≤ 0.05) correlated with amounts of these metals taken up by corn plants (Table 11). In the of wheat plants, the significant correlations were found only for Zn and Cu in

straw and Ni in wheat seeds (Table 12). The presence of acetate anions with EDTA enhanced its power for extracting the available metals. The amounts of available Cd, Ni, and Cr extracted with EDTA-AAc extraction solution significantly $(p \le 0.05)$ correlated with the amounts of metals taken up by

corn plants, and the amounts of available Fe, Mn, Zn, Cu, and Pb with EDTA-AAAc extracted extraction solution significantly $(p \le 0.05)$ correlated with the amounts of metals taken up by corn plants. With wheat plants, significant ($p \le 0.05$) correlations were obtained only for Zn and Pb in straw using EDTA-AAc, and Mn, Cu, and Pb in straw using EDTA-AAAc (Table 12). The extractable metals identified by **EDTA** methods exhibited relatively poor or no correlation with the other metal content of wheat plants. This discrepancies in the results emphasized that the EDTA extraction procedures specific for all were not micronutrients and heavy metal extractions. The presence acetate anions in the extraction solutions enhanced the potential extracting power of the solution, and resulted in extracting high amounts of metals from different pools, including the occluded and poorly crystalline forms. Thus, the amounts of metals extracted by these procedures were either weekly correlated or did not correlated at all with the amounts of metals uptake by plants. These results lead to the general conclusion that no one extraction procedures is suitable for testing the status of all metals in all soils after all crops. Precautions should be taken when deciding which extraction solution should be used.

References

- Abd El-Haleem A.A., H.M. Salem, R.S. El-Aal, and E. Schnug. 2002. Extractability and availability of cationic micronutrients in soils of Egypt. Annals of Agric. Sc., Moshtohor, Vol. 40(1): 683-698.
- Abdel Salam, A. A. 2002. Distribution of heavy metals in three soil types and in corn plants at western area of Nile delta. J. Agric. Sci. Mansoura Univ., 27 (12): 8713-8734.
- Bermond, A., I. Yousfi, and J.P. Ghestem. 1998. Kinetic approach to the chemical speciation of trace metals in soils. *Analyst 123, 785-789*.
- Brazauskine D.M., N. Sabiené, and D. Rimmer. 2004. Determination of heavy metal mobile forms using different extraction methods. Ekologija. Nr.1. p. 36-41.
- Bruemmer, G.W., J. Gerth, and U. Herms. 1986. Heavy metal species, mobility and availability in soils. J. Plant Nutrition and Soil Sci. Vol. 149: 382-398.
- Chapman, H.D. and P.E. Pratt. 1961. Method of Analysis for Soils, Plants and Waters. Univ. California; *Diusion of* Agric. Science.
- Cox, F.R. and E.J. Kamprath. 1972. Micronutrient Soil Tests. In Mortvedt et al. (Eds.) Micronutrients in Agriculture. 1972. Soil Sci.

- Soc. Am, Inc. Madison, Wisconsin USA. pp289-317.
- El-Koumey, B.Y., F. Abu Agwa, and M.F. Tantawy. 1997: Ni, Pb and Cd uptake by plant as a function of their extractability from different soils. *Menofiya J. Agric. Res. Vol. 22. No 5:1411-1430.*
- El-Sokkary, H. and A.I. Sharaf. 1996. Enrichment of soils and plants irrigated by wastewater by zinc and cadmium. Egypt. J. Soil Sci., 36(1-4): 219-232.
- Feng, M.H., X.Q. Shan, S.Z. Zhang, and B. Wen. 2005. Comparison of a rhizosphere-based method with other one-step extraction methods for assessing the bioavailability of soil metals to wheat. Chemosphere 59: 939-949.
- Gupta, A.K., and S. Sinha. 2006. Chemical fractionation and heavy metal accumulation in the plant of Sesamum indicum (L.) var. T55 grown on soil amended with tannery sludge: Selection of single extractants. Chemosphere 64: 161-173.
- Hegazy, M.N.A. 1980. Studies on iron availability in soils. M.Sc. Thesis. Fac. Agric., Cairo, University.
- Hegazy, M.N.A., E.A. El-Sayed, and A.M. Sorour. 1991. Evaluation of some extractants for determining available Fe, Mn, Zn and Cu in some Egyptian soils. Zagazig J. Agric. Res., 18: 897-910.

- Kabata-Pendias, A., and H. Pendias. 1984.Trace elements in soils and plants. CRC press inc. Boca Raton. Florida, 33431.
- Kabata-Pendias, A., and H. Pendias. 1992. Trace elements in soils and plants. 2nd ed. CRC Press, Boca Raton, Florida.
- Lindsay, W. L. and W. A. Norvell. 1978. Development of a DTPA soil test for zinc, iron, manganese, and copper. Soil Sci. Soc. Amer. J. 42:421-428.
- Melsted, S.W. and T.R. Peck. 1973. The Principles of Soil Testing-2. In Walsh, L.M. and J.D. Beaton (Ed.) Soil Testing and Plant Analysis. 1973. Soil Sci. Soc. Am, Inc. Madison, Wisconsin USA.
- Nomeda, S., D.M. Brazauskienë, and D. Rimmer. 2004. Determination of heavy metal mobile forms by different extraction methods. *Ekologija*. Nr. 1. P. 36-41.
- Nyamangara, J., and J. Mzezewa. 1999. The effect of long-term sewage sludge application on Zn, Cu, Ni and Pb levels in a clay loam soil under pasture grass in Zimbabwe. Agric. Ecosys. Environ. 73, 199-204.
- Parkinson, J.A. and S.E. Allen. 1975. A wet oxidation procedure for the determination of N and mineral nutrients in biological materials. Comm. Soil Sci. Plant Anal. 6:1-11.

Podlesakova, E., J. Nemecek and R. V'acha. 2001. Mobility and Bioavailability of trace elements in soil in "Trace Elements in soils. Bioavailability, Flux, and Transfer, Iskandar, I.K. and M.b. Kirkham Lewis Publishess C.R.C. pp21-41.

Trierweiler, and Lindsay. 1968. EDTA- Ammonium carbonate soil test for zinc. Soil Sci. Soc. Amer. Proc., vol. 33:49-53.

Ure, A.M. 1996. Single extraction schemes for soil analysis and related applications. The science of the total environment. Vol. 178. P. 3-10.

Vose, P. B., and P. J. Randall. 1962. Resistance to aluminum and manganese toxicities in plants related in variety and cation-exchange capacity. Nature 196: 85.

Wear, J.I. and Evans, J.G. 1968. Relationship of Zn uptake by corn and sorghum to soil Zn measured by three extractants. Soil Sci. Soc. Am. Proc. 32: 543-546.

Zhu, B. and Alva, A.K. 1993. The chemical forms of Zn and Cu extractable by Mehlich 1, Mehlich 3, and ammonium bicarbonate-DTPA extraction. Soil. Scie. Vol.156, No.4: 251-258.

قابلية الاستخلاص و التيسر لبعض العناصر المعدنية في الأراضي الملوثة بأسيوط.*

شيماء مصطفى كامل

قسم الأراضي والمياه - كلية الزراعة - جامعة أسيوط

تم إجراء هذه الدراسة بهدف تحديد انسب المحاليل التي يمكن استخدامها لاستخلاص بعض العناصر الصغرى و العناصر الثقيلة في الأراضي الملوثة بمياه المجارى.

و قد تم أخذ 18 قطاع تمثل 6 قرى تروى بمياه المجارى الملوثة من أكثر مــن 45 عام بمحافظة أسيوط حيث أخذت عينات سطحية و تحت سطحية. كمــا زرعــت هــذه الأراضي بنباتي الذرة و القمح لربط المحتوى العنصري لمهذه النباتات بمحتوى التربة من نفس العناصر (الحديد – المنجنيز – الزنك – النحاس – الكادميوم الرصاص – النيكل – الكروم).

و من هذه الدراسة أمكن التوصل إلى كثير من النتائج الهامــة مــع تحديــد أنســب المستخلصات التي يمكن استخدامها في مثل هذه النوعية من الأراضي (الأراضي الملوثة فقط) .

كما توصى هذه الدراسة إلى التوسع في هذه النوعية مــن الأبحـــاث لتشـــمل بقيـــة الأراضي الغير ملوثة بأنواعها المختلفة.

^{*} بحث مقدم إلى المؤتمر العلمي الثاني الشباب الباحثين بكلية الزراعة جامعة أسيوط، 6 مايو 2008.