

## Role of Plant Antioxidants (Ascorbic Acid, Glutathione, $\alpha$ -Tocopherol, Spermine) in Alleviating Salinity Stress on Growth and Yield of Wheat Plants

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**A**POT experiment was carried out to investigate the role of some plant antioxidants such as ascorbic acid, glutathione,  $\alpha$ -tocopherol, spermine in alleviating the harmful effects caused by high salinity level (6g/l) in wheat plants. The seeds were pre-soaked and the plants sprayed with anyone of the antioxidants used. It was concluded that salinity stress depressed all growth parameters and yield components. The data also showed that the different antioxidants could partially alleviate the harmful effect of salinity stress on growth and yield of wheat plants.

The data showed that 6000 mg/l salinity level alone or in combination with any of antioxidants used increased the activity of total peroxidase, ascorbic peroxidase, superoxide dismutase and catalase in wheat leaves. In addition, salinity (6000 mg/l) alone or in combination with any of applied antioxidants increased the endogenous contents of ascorbic and glutathione and total phenols but decreased carotenoids.

**Keywords:** Wheat, Plant antioxidant, Salinity stress, Growth and yield.

Soil salinity is one of the major abiotic stresses affecting crop growth and productivity. Salt stress causes inhibition of growth and development, reduction in photosynthesis, respiration, and protein synthesis and disturbs nucleic acid metabolism. Decrease in uptake of  $K^+$ ,  $Mg^{2+}$ ,  $Ca^{2+}$ , and thereby reduced growth at higher sodium concentration have also been reported (Sairam & Srivastava, 2002). Yang *et. al.* (1990) reported that there are two ways that salinity could retard growth: (a) By damaging the growth of cells so that they can not perform their functions or (b) By limiting their supply of essential metabolites. Sakr (1996) indicated that salinity suppressed both cell division and cell enlargement proportionally in wheat plants. Regarding the effect of antioxidant on wheat under salinity stress, Shalata & Neumann (2001) found that ascorbic acid acts directly to neutralize superoxide radicals, single oxygen or superoxide and as a

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secondary antioxidant during reductive recycling of the oxidized form of  $\alpha$ -tocopherole.

It is now widely accepted that reactive oxygen species (ROS) are responsible for various stress-induced damage to macromolecules and ultimately to cellular structure. Consequently the role of antioxidant enzymes viz. superoxide dismutase (SOD), ascorbate peroxidase (APX), glutathione reductase (GR) and catalase (CAT), as well as metabolites like ascorbic acid, glutathione,  $\alpha$ -tocopherol, flavonoids, carotenoids (CAR) are responsible for the quenching of ROS becomes very important (Sairam & Srivastava, 2002).

Ascorbic acid (AA) is an important antioxidant, which reacts not only with  $H_2O_2$  but also with  $O_2$ , OH and lipid hydroperoxidases. On the other hand, AA has been associated with several types of biological activities in plants: (1) As an enzyme co-factor, (2) As an antioxidant and (3) As a donor/ acceptor in electron transport at the plasma membrane or in the chloroplasts, all of which are related to oxidative stress resistance (Conklin, 2001).

Regarding enzymatic antioxidants activity caused by salinity stress, Vaidyanathan *et al.* (2003) reported that the non-enzymatic antioxidants (glutathione and ascorbate) showed an accumulation in root tissues in plants subjected to salt stress. Plants must possess efficient antioxidant system such as SOD, APX, GR, DIAR, CAT and metabolites viz., ascorbic acid, glutathione,  $\alpha$ -tocopherol, carotenoid, flavanoids etc. (Smirnoff, 1995).

Regarding the effect of polyamines, Smith (1975) established a specific role of polyamines in maintaining a cation-anion balance in plant tissues and in stabilizing membranes at high external salinity. Polyamines had an ameliorating effect on all morphological and physiological characters and prevented degradation of chlorophyll, however, polyamines enhanced accumulation of all organic compounds under salinity stress study, except phenols (Zhao & Qin, 2004).

The objective of this study was to investigate the role of selected antioxidants on mitigation or alleviation the harmful effect of salinity stress on wheat plant.

### Materials and Methods

A pot experiment was carried out at the seed technology unit of Tag El-Ezz Research Station in Dakahlia Governorate, Agricultural Research Centre, Ministry of Agriculture, Egypt, during the 2006/2007 season. Wheat grains (var. Gemeza 9) were kindly supplied by the Plant Breeding Section of the Field Agricultural Research Centre, Ministry of Agriculture, Giza, Egypt. Uniform grains were soaked in antioxidants *i.e.* ascorbic acid (100 mg/l), reduced glutathione (100 mg/l),  $\alpha$ -tocopherol (50 mg/l), spermine (10 mg/l) as well as tap water for 6 hr. Eight uniform grains were sown on November 15<sup>th</sup>, 2006, in plastic pots (30 cm diameter) under high salinity level (6000 mg/l NaCl). The plants were sprayed with the same antioxidant three times (30, 60 and 90 days from sowing).  
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Growth parameters were measured in the vegetative phase (45 days after sowing). At physiological maturity yield and its components were recorded.

All the normal culture practices of growing wheat plants were applied as usual manner followed by the farmer in the district.

Leaves were collected after 75 days from sowing then frozen and stored till analysis. Each treatment was replicated 3 times and arranged in a complete randomized block design (Araus *et al.*, 2008).

#### *Biochemical constituents*

##### *Non-enzymatic antioxidant contents*

*Ascorbic acid determination:* 0.5 g of fresh leaves was ground in 50 ml of 2% (w/v) metaphosphoric acid using mortar and pastel and centrifuged for 30 min at 13000 rpm at 4°C. The ascorbate content ( $\mu\text{mol} / \text{g FW}$ ) was measured in the supernatant at 25°C. The absorbance of red colour was measured at 520 nm according to Omaye *et al.* (1979) and Athar *et al.* (2008).

*Total glutathione determination:* The level of total glutathione (GSH) was determined in the fresh leaves with Ellman's reagent according to De Vos *et al.* (1992). 300  $\mu\text{l}$  of sample buffer were mixed with 630  $\mu\text{l}$  of 0.5 M  $\text{K}_2\text{HPO}_4$  and 25  $\mu\text{l}$  of mM 5, 5 -dithiobis (2-nitrobenzoic acid) (final pH 7). The absorbance at 412 nm was read after 2 min. GSH was used as a standard.

*Total phenols determination:* 1g of the fresh leaves were macerated in 5- 10 ml 80% ethanol for at least 24 hr at 0°C, the alcohol was clarified, the remained residue was re-extracted with 5-10 ml 80% ethanol 3 times. At the end, the clarified extract was completed to 50 ml using 80% ethanol. The colorimetric method of Folin-Denis as described by Daniel & George (1972) was employed for the chemical determination of phenolic compounds. Quantities were determined by reading the developed blue colour at 725 nm using 0.5 ml 80% ethanol and reagents only as a blank.

*Carotenoids determination:* Leaf samples (0.5gm from the third upper foliage leaf) were subjected to extraction by methanol for 24 hr at lab. temp. after adding a trace from sodium carbonate (Robinson *et al.*, 1983); then, carotenoids were determined spectrophotometrically (Spekol11) at wave-length 452 and calculated by the equation introduced by Mackinny (1941).

##### *Enzymatic antioxidant activity determination*

*Ascorbate peroxidase activity determination:* Ascorbate peroxidase (As-POD) was assayed spectrophotometrically according to Fielding (1978). The assay was carried out at 25°C in 1.0 cm light path cuvette and the reaction mixture consisted of 1500  $\mu\text{l}$  phosphate buffer, 20  $\mu\text{l}$  EDTA, 1000  $\mu\text{l}$  sodium ascorbate and enzyme extract (20  $\mu\text{l}$ ). After mixing, the reaction was initiated by adding the 480  $\mu\text{l}$   $\text{H}_2\text{O}_2$  and decreasing in optical density at 290 nm against blank (without extract) was continuously recorded every minute (for two minutes).

*Total peroxidases activity determination:* Peroxidase was assayed spectrophotometrically according to Amako *et al.* (1994). The assay was carried out at 25°C in 1.0 cm light path cuvette and the reaction mixture consisted of 1500 µl phosphate buffer, 1000 µl Pyrogallol and 480 µl H<sub>2</sub>O<sub>2</sub> solution. After mixing the reaction was initiated by adding the enzyme extract (20 µl) and the increase in optical density at 430 nm against blank (without extract) was continuously recorded every minute (for three minutes).

*Super oxide dismutase (SOD) activity determination:* Leaf samples were collected in an ice bucket and brought to the laboratory. Leaves were then washed with distilled water and surface moisture was wiped out. Leaf samples (0.5g) were homogenized in ice cold 0.1M phosphate buffer (pH 7.5) containing 0.5Mm EDTA with pre-chilled pestle and mortar. The homogenate was transferred to centrifuge tubes and was centrifuged at 4°C in Beckman refrigerated centrifuge for 15 min at 15000 X g. The supernatant was transferred to 30 ml tubes and referred to enzyme extract.

SOD activity was estimated by recording the decrease in absorbance of superoxide-nitro blue tetrazolium complex by the enzyme. About 3ml of reaction mixture containing 0.1ml of 1.5M sodium carbonate, 0.2ml of 200mM methionine, 0.1ml of 2.25Mm Nitro-blue tetrazolium, 0.1ml of 3mM EDTA, 1.5ml of 100mM potassium phosphate buffer, 1ml distilled water and 0.05 ml of enzyme were taken in test tubes in duplicate from each enzyme sample. Two tubes without enzyme extract were taken as control. The reaction was started by adding 0.1ml riboflavin (60mM) and placing the tubes below a light source of two 15 w florescent lamps for 15 min. Reaction was stopped by switching off the light and covering the tube with black cloth. Tubes without enzyme developed maximal colour. A non- irradiated complete reaction mixture which did not develop colour served as blank. Absorbance was recorded at 560nm and one unit of enzyme activity was taken as the quantity of enzyme which reduced the absorbance reading of samples 50% when comparison with types lacking enzymes (Dhindsa *et al.*, 1981).

Catalase activity was determined by measuring the rate of H<sub>2</sub>O<sub>2</sub> conversion to O<sub>2</sub> at room temperature using a lipid-phase O<sub>2</sub> electrode (Hansatech, Norfolk, UK). Approximately 0.5g of plant tissue, consisting of the apical region of the shoot including the cotyledons, was extracted in 1.5 ml of 0.1M Hepes/KOH - buffer (pH 7.4) and then centrifuged at 10.000g for 5 min. The rate of O<sub>2</sub> production was measured by adding 50 µl of the supernatant to 0.1M Hepes (pH 7.4) containing 530mM H<sub>2</sub>O<sub>2</sub>. Catalase activity was calculated on a fresh weight basis to keep the data uniform with the H<sub>2</sub>O<sub>2</sub> measurements, and to reduce the chances of distribution as a result of protein synthesis alteration due to a heat-shock (Vierling, 1991 and Bettany, 1995).

The data of experiments were statistically analyzed as technique of the analysis of variance (ANOVA) according to Gomez & Gomez (1984). The treatment means were compared using the least significant differences (LSD).

**Results**

*Plant growth*

The data presented in Table 1 show that a high salinity level (6000 mg/l) alone or in combination with any of applied antioxidants (ascorbic acid, glutathione,  $\alpha$ -tocopherol, spermine) decreased stem and leaf dry weights. Different applied antioxidant materials could partially counteract the harmful effect of high salinity stress level on growth of wheat plant.

**TABLE 1. Effect of antioxidants on growth, yield, endogenous enzymatic and non-enzymatic antioxidants in wheat plants grown under a high salinity level (6 g/l). Combined analysis of the two growing seasons.**

Parameters Treatments	Stem DW (g)	Leaf DW (g)	Spike length (cm)	No. spikes/ plant	Spike weight/ plant (g)	No. grains/ spike	Grains weight/ spike (g)	Grains yield/ plant (g)
Tap water (control)	4.1	4.5	10.0	40	2.5	36	2.5	14.7
Salinity 6 g/l	2.4	3.1	7.3	23	1.6	28	1.6	10.2
S + ASA. 100mg/l	3.3	4.3	9.6	38	2.0	37	2.3	13.6
S + Glut. 100mg/l	3.1	4.3	8.7	35	1.9	35	2.2	13.6
S +Tocoph. 50mg/l	4.0	4.2	8.6	32	1.7	36	2.0	13.0
S + Sper. 10mg/l	3.2	4.0	8.3	33	1.6	35	2.2	13.0
LSD at 5%	1.9	0.43	0.31	0.99	0.16	2.6	0.16	0.13
	<b>TPX activity</b>	<b>APX activity</b>	<b>Cat activity</b>	<b>SOD activity</b>	<b>Carot. content</b>	<b>ASA content</b>	<b>GSH content</b>	<b>Total phenols</b>
Tap water (control)	152	125	53	307	0.48	7.9	220	129
Salinity 6 g/l	193	141	63	382	0.15	8.9	246	236
S + ASA 100 mg/l	386	218	163	380	0.28	10.2	291	286
S + Glut. 100 mg/l	341	154	165	396	0.21	10.1	206	274
S +Tocoph. 50 mg/l	223	157	106	366	0.17	9.6	278	233
S + Sper. 10 mg/l	276	218	102	360	0.32	9.1	235	287
LSD at 5%	1.78	2.67	1.52	1.62	0.076	0.18	3.04	1.36

TPX: total peroxidase activity (units /g fresh weight). APX: ascorbic peroxidase activity (units /g fresh weight). SOD: superoxide dismutase activity (units /mg protein/min). Cat: catalase activity  $\mu\text{mol H}_2\text{O}_2$  red/mg protein/min). Carot.: carotenoids content (mg/g fresh weight). ASA: ascorbic acid content (mg/g fresh weight). GSH: reduced glutathione ( $\mu\text{mol /g fresh weight}$ ). T. phenols: total phenols content (mg/g fresh weight). S: salinity stress at 6000mg/l. ASA: ascorbic acid. Glut.: glutathione. Tocoph: tocopherol. Sper.: spermine.

*Yield and its components*

A high salinity level (6000 mg/l) significantly decreased yield and its components as spike length, number of spikes/plant, spike weight, number of grain/spike, grain weight/spike and grains yield/plant. A high salinity level combined with any of applied antioxidants also decreased wheat yield and its components. The applied antioxidants mitigated the effects of salinity. It was noticed that the antioxidants can partially alleviate the harmful effects of salinity

stress on yield and its components. Glutathione and spermine showed the most beneficial effect in this respect.

#### *Enzymatic antioxidants activity*

The data show that 6000 mg/l salinity level alone or in combination with any of antioxidants used increased enzymatic activity of total peroxidase (TPX), ascorbic peroxidase (APX), super oxide dismutase (SOD) and catalase. High salinity level combined with ascorbate proved to be more effective in increasing all enzymatic antioxidants activity. Moreover the combination of glutathion is more effective in increasing catalase and superoxide dismutase activity, while spermine was more effective in increasing total peroxidase and ascorbic peroxidase activity.

#### *Non enzymatic antioxidant contents*

The data show that high salinity (6000mg/l) alone or combined with any of applied antioxidants decreased carotenoids content, while increasing each of the endogenous ascorbic, glutathione or total phenol contents. Ascorbic treatment was the most effective in improving endogenous non-enzymatic antioxidants.

It could be noticed that the combination between high salinity level and any of exogenous applied antioxidants caused synergistic effect on increasing endogenous non-enzymatic antioxidants such as ascorbic, glutathion and total phenols.

### **Discussion**

#### *Growth and yield*

The inhibitory effect of salinity on wheat growth in the present investigation may be due to a decrease in water absorption, metabolic processes, meristematic activity and/or cell enlargement (Khadr *et al.*, 1994 and Sakr, 1996). Yang *et al.*, (1990) reported that there are two ways that salinity could retard growth: (a) By damaging growth cells so that they can not perform their functions, (b) By limiting their supply of essential metabolites. Perturbing the functioning of vital components of photosynthesis like PSI, PSII and Rubisco, (Chen & Murata, 2002 and Stepien & Klobus, 2005).

Salinity affected all stages of wheat growth and development, as well as yield of plants. The yield was much more depressed by salt than the vegetative growth which may be attributed to a decrease in the viability of pollen grain or in the receptivity of the stigmatic surface or both (Sakr *et al.*, 2004).

The reduction in seed yield is largely due to : (1) Reduction in pollen viability which has been related to decreased calcium mobilization from plant leaves treated with sodium chloride, which is important in pollen germination and pollen tube growth (Grattan *et al.*, 2002). (2) Abscission of flowers or young fruit due to ethylene induction by salinity (Roy *et al.*, 1995). (3) Decreasing production pollen grain, mean number of perfect flowers, and fruit set. (4) Decreasing the leaf area and number per plant, resulting reduction in the supply of carbon assimilate due to decreasing the net photosynthetic rate and biomass accumulation (Sakr *et al.*, 2004).

As for,  $\alpha$ -tocopherol, ascorbic acid and glutathione, it could be concluded that, these plant antioxidants can alleviate the harmful effect of ROS caused by salinity stress through several ways such as: (1) Inhibiting the lipid photoperoxidation (Michalski & Kaniuga, 1981). (2) Involving in both electron transport of PS II and antioxidizing system of chloroplasts (McKersie & Leshem, 1999). (3) Membrane stabilizers and multifaceted antioxidants, that scavenge oxygen free radicals, lipid peroxy radicals and singlet oxygen (Diplock *et al.*, 1989). (4) Reacting with peroxy radicals formed in the bilayer as they diffuse to the aqueous phase (Hess, 1993). (5) Scavenging cytotoxic  $H_2O_2$ , and reacts non-enzymatically with other ROS: singlet oxygen, superoxide radical and hydroxyl radical (Larson, 1988). (6) Regenerating another powerful water-soluble antioxidant, ascorbic acid, via the ascorbate–glutathione cycle (Blokhina *et al.*, 2002). (7) Stabilize membrane structures (Blokhina, 2002). (8) Modulating membrane fluidity in a similar manner to cholesterol, and also membrane permeability to small ions and molecules (Foyer, 1992). (9) Decreasing the permeability of digalactosyl-diacyl-glycerol vesicles for glucose and protons (Berghlund *et al.*, 1999).

Regarding polyamines (PAs), it has been suggested that PAs may play role in anti-oxidative system and protect membrane from peroxidation. The alleviating effect of polyamines on plants grown under salinity stress may be due to one or more of the following factors: (1) Through activating anti-oxidative defense system (Chattopadhyay *et al.*, 2002). (2) Suppressed the level of superoxide and  $H_2O_2$  in leaf stressed plants (Hernandes *et al.*, 1995). (3) Suppress  $H_2O_2$  level and thereby membrane damage is being evaluated in terms of anti-oxidative system (Dionisio–Sese & TOBITA, 1998). (4) Caused reduction in ROS through quenching of singlet oxygen and excited chlorophyll by elevating level of CAR thereby maintained chloroplastic membrane (Velikova *et al.*, 2000). (5) reduce membrane leakage and lipid peroxidation and decreased MDA contents in sugarcane leaves (Zhang & Kirkham, 1996). (6) Stabilization of membrane damage may be due to its polycationic nature (Tiburcio *et al.*, 1994). (7) Increasing AXP and GR activity as well as CAR and GSH at all salinity levels (Tiburcio *et al.*, 1994). (8) Stimulation of chlorophylls synthesis and prevent chlorophyll degradation (Krishnamurthy, 1991). (9) Increasing all organic concentrations, that may be attributed to that polyamines that are involved in important biological processes, *e. g.* ionic balance and DNA, RNA and protein synthesis .

Generally speaking, the application of antioxidants proved to be more effective in alleviating the harmful effect of salinity on wheat plant. However, ascorbic and spermine showed the most beneficial effect in this respect (Sakr *et al.*, 2004).

Regarding enzymatic antioxidant activity, Dash & Panda (2001) reported that higher activity of antioxidant enzymes viz., SOD, GR and CAT caused lower  $H_2O_2$  production, lipid peroxidation and higher membrane stability. Beneficial effect of higher osmolyte concentration (soluble sugars, glycine-betaine, proline and potassium) is reflected in stabilization of essential enzyme proteins such as SOD, GR and CAT resulting in higher activity under salinity stress.

Plants possess antioxidant systems in the form of enzymes such as SOD, APX, GR, DIAR, CAT and metabolites viz., ascorbic acid, glutathione,  $\alpha$ -tocopherol, carotenoid, flavanoids, etc. These antioxidant enzymes and metabolites are reported to increase under various environmental stresses as well as comparatively higher activity has been reported in tolerant cultivars than the susceptible ones (Sairam & Srivastava, 2002).

As for non enzymatic antioxidants, ascorbic acid has been associated with several types of biological activities in plants: (1) As an enzyme co-factor, (2) As an antioxidant, and (3) As a donor/ acceptor in electron transport at the plasma membrane or in the chloroplasts, all of which are related to oxidative stress resistance (Conklin, 2001).

Total phenols in the wheat leaves tended to increase gradually with increasing salinity levels in soil as shown in Table 1. This increase showed some tendency of wheat to adjust osmotically against salt stress. Moreover, stress condition leads to an increase in phenolic compounds (Namiki, 1990). These phenolic compounds could be a cellular adaptive mechanism for scavenging oxygen free radicals during stress and this free radical scavenger and others such as ascorbate could be readily oxidized in the system of tissue representing subcellular damages.

Reducing glutathione (GSH) can function as an antioxidant in many ways. (1) It can react chemically with singlet oxygen, superoxide and hydroxyl radicals and therefore function directly as a free radical scavenger (Noctor & Foyer, 1998). (2) GSH may stabilise membrane structure by removing acyl-peroxides formed by lipid peroxidation reactions. (3) GSH is the reducing agent that recycles ascorbic acid from its oxidised to its reduced form by the enzyme dehydroascorbate reductase (Loewus, 1988).

The increase in GSH contents as reported in Table 1 may be due to the role of the enzymes ascorbate peroxidase, glutathione reductase, superoxide dismutase which involved in the regeneration of glutathione and ascorbate that are important in detoxification of ROS (Foyer *et al.*, 1994).

Generally, it was concluded that tocopherol, ascorbic and glutathione can help to alleviate the harmful effect of ROS may be through several ways such as: (1) Inhibiting the lipid photoperoxidation (Michalski & Kaniuga, 1981). (2) Affecting, both electron transport of PS II and the antioxidizing system of chloroplasts (McKersie & Leshman, 1999). (3) Stabilizing membranes (Thomas *et al.*, 1992). (4) Reacting with peroxy radicals (Sairam & Srivastava, 2002). (5) Scavenging cytotoxic  $H_2O_2$  and reacting non-enzymatically with other ROS (Larson, 1988). (6) Regenerating another powerful water-soluble antioxidant, ascorbic acid (Blokhina *et al.*, 2002). (7) Stabilizing membrane structures (Blokhina *et al.* 2002). (8) Modulating membrane fluidity in a similar manner to cholesterol and also membrane permeability to small ions and molecules (Foyer, 1992) and (9) Decreasing the permeability of digalactosyldiacylglycerol vesicles for glucose and protons (Berglund *et al.*, 1999).

As for the role of exogenous antioxidant polyamine on alleviating salinity stress in plants, the polyamines (PAs) including spermine (SPM, a tetramine), spermidine (SPD, a triamine) and their obligate precursor putrescine (a diamine) are implicated in induction of plant adaptation to stresses (Mishra *et al.*, 2003). It was suggested that PAs may play role in antioxidative system and protect membrane from peroxidation. The alleviating effect of polyamines on plants grown under salinity stress was previously discussed.

From the above-mentioned reported results: it could be noticed that the applied antioxidants could alleviate or minimize the harmful effect of NaCl salinity on wheat plant and ascorbic and spermine proved to be more effective in this respect.

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## دور منظمات النمو (حمض الأسكوربيك والجلاتثيون والتوكوفيرول والإسبرمين) فى تخفيف الإجهاد الملحي على نمو ومحصول نبات القمح

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أجريت تجربة أصص لدراسة دور بعض مضادات الأكسدة مثل حمض الأسكوربيك والجلاتثيون والتوكوفيرول والإسبرمين لتخفيف الأثر الضار المتسبب عن مستوى الملوحة المرتفع (6 جم/لتر) لنباتات القمح. تم نقع البذور ورش النباتات بمضادات الأكسدة وتبين أن الملوحة أدت إلى تقليل القياسات الخضرية والمحصولية. وكذلك أوضحت النتائج دور مضادات الأكسدة فى تقليل الأثر الضار للملوحة فى كل من القياسات الخضرية والمحصولية لنبات القمح.

أوضحت البيانات أن تركيز الملوحة 6000 مليجرام/لتر منفردا أو مع أى من مضادات الأكسدة زاد من نشاط البيروكسيداز الكلى وأنزيم الأسكوربيك بيروكسيداز والسوبر أكسيداز ديسميتيز والكتاليز فى أوراق القمح. بالإضافة إلى زيادة المحتوى الداخلى لكل من الأسكوربيك والجلاتثيون والفينولات الكلية، لكن أنقصت الكاروتينيدات.