

## OPTIMAL USE OF COMBINED SOIL AMENDMENTS FOR RECLAIMING AND RESTORING SALT-AFFECTED SOILS

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### ABSTRACT

A leaching column experiment techniques was carried out to evaluate the optimal combination of gypsum, sulfuric acid, limestone, and organic compost as soil reclamation amendments for a saline-sodic soil taken from Sahl El Tina, Egypt. After termination of leaching, the soil columns were planted with the halophyte plant, *Atriplex halimus*. Leaching as well as irrigation was done using El-Salam canal water of  $1.72 \text{ dSm}^{-1}$ . All of the amendments, singly or in combination decreased EC, pH, SAR and ESP soil properties and increased exchangeable calcium. Initial values being  $28.8 \text{ dSm}^{-1}$ , 8.40, 12.96 and 38.01 for soil EC, pH, SAR and ESP, respectively, decreased upon leaching to 5.80, 7.99, 6.05 and 12.40; then to 2.10, 7.7, 4.06 and 8.80 after harvesting, respectively, as general means. The obtained data showed that there is no one combination fitting for all soil properties of EC, pH, SAR, ESP and bulk density although using a mixture of the 4 amendments instead of one kind may be recommended. Growing the halophyte plant increased the effectiveness of amendments.

**Keywords:** Reclamation, Gypsum, Sulfuric acid, Limestone, Compost, Computer model.

### INTRODUCTION

Reclaiming saline and particularly sodic soil is a recurring and challenging problem. A general rule was to apply a given depth of water to remove

80% of the soluble salt from the same depth of the soil. Early, Carter and Robbins (1978) reported that essentially all of the residual salt could be removed from soil profiles by passing 30 cm of water per meter depth of soil

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profile. However, removal of exchangeable sodium necessitates application of chemical amendments to remove the sodium from the soil's cation exchange sites. Sahin *et al.* (2002) proved that reclamation of saline sodic and sodic soils generally starts by increasing calcium on the exchange complex at the expense of sodium. Calcium ions required for the exchange reaction could be obtained either from the added calcium amendments or from the native calcium carbonate which could be mobilized through the addition of acids or acid formers (Kamphorst and Aolt, 1976; Singh *et al.*, 1981; Bresler *et al.*, 1982; Loveday, 1984 and Abou Yuossef 2001), Loveday (1984) and Yahia *et al.*, (1975) demonstrated the effectiveness of surface-applied sulfuric acid for reclaiming sodic soils and reported that applications of 5 to 15 Mg/ha proved superior to surface-applied gypsum for calcareous sodic soils. Prather *et al.* (1978) summarized the advantages of sulfuric acid, gypsum, and  $\text{CaCl}_2$  singly or in combination for sodic soil reclamation. Their results indicate that a combination of amendments may provide effective in reclamation at low costs. On the other hand various organic

amendments such as mulch, manures, and compost have been investigated for their effectiveness on remediation of saline-sodic soils (Diez and Krauss 1997; Wahid *et al.*, 1998). In general, organic amendments have a very little effect on improving soil salinity and sodicity when they are applied alone (Madejon *et al.*, 2001). However their effectiveness in improving many soil properties is well documented in literature (Cheny and Swift, 1984; Hanay *et al.*, 1992; Gao and Chang, 1996; Prihar *et al.*, 1996; Entry *et al.*, 1997; Giusquiani *et al.*, 1995; Ibrahim and Shindo, 1999; Mamo *et al.*, 2000; Nacini and Cook, 2000). Al-Khateeb *et al.*, (2001) concluded that planting halophyte plants in saline soils would help in accumulating sodium in leaves and reclamation of saline soils. The current work was undertaken to predict the optimal combination of some soil amendments for saline sodic reclamation using a tetra factorial computer model.

## MATERIALS AND METHODS

A tetra factorial computer model by Moussa and Youssef (1992) was applied to assess the effect of using gypsum "G", limestone "L", sulfuric acid "S" and

organic compost "C", on chemical properties of the soil as shown in Tables 1 and 2. The mathematical approach and elevation model could be found in Chung (2002). Application of gypsum based on the gypsum requirements (GR) equation (USDA, 1954) taking in consideration a required final value of exchangeable sodium percent ( $ESP_f$ ) to be 10%, the actual exchangeable sodium percent ( $ESP_i$ ) is equal 38.01% (Table 1). The gypsum was of 98.81% purity and its addition rate was 13.5  $Mg\text{fed}^{-1}$  (which costs 2029.53 LE). The investigated materials in the experiment (G, L, S, and C) were designated as X1, X2, X3, and X4 respectively. However the levels of amendment ranging from zero to a maximum. The maximum dose of each amendment is chosen to so as to cost (i.e. in terms of cost rather than gypsum requirement) 2029.53 LE / feddan, consequently values for the maximum dose for G, L, S and C in terms of material quantity are 13.5, 5.80, 0.25 and 8.12  $Mg\text{fed}^{-1}$ , respectively.

There were 19 treatments, which cover all the possible combination of the amendments (Table 3). Following the above mentioned Tetra-factorial model, the four amendments G, L, S and C were allocated the four heads of

the tetrahedron on each top site they were equal to maximum 100% or 8 points graduated to be 0% on the opposite base. Treatments 1 to 4 and 6 to 11 lie on the surface of the tetrahedron while treatments 12 to 19 lie inside it, whereas treatment 5 lies exactly on the tetrahedron center which consists of the four amendments (25% each) or 2 units each (the sum of any treatment is 100% or 8 units, with equal the fixed cost of 2029.53 LE/fed.. Treatments were done in 3 replicate each in plastic cylinder columns of 75 cm height and 16 cm inside diameter. The bottom of each column was sealed with perforated mesh nylon screen and glass wool. Acid washed inert sand (pre-washed with HCl and water) was placed on the tube bottom to make a 5 cm layer of the column to regulate the flow of water and to prevent plugging the lower part of column by the immigrating fine materials. Soil was packed in the tubes so as to a soil column 60 cm height and bulk density of 1.65  $Mg\text{m}^{-3}$ . This required a quantity of soil of 20 kg of air-dried soil per column. Such arrangement allowed for 10 cm on top of soil column to give sufficient space for addition of water for irrigation and leaching process. Treatments were prepared

according to Table 3. Amendments were mixed homogeneously within the top 30 cm to be ready for starting the leaching process.

Water taken from El-Sallam Canal (used irrigate Sahl El-Tina) was used for the leaching and irrigation process (Table 1). The leaching procedure using the intermittent method was done so as to add water portions to the already saturated soil columns; and obtain leachates which are equal to the added portions. The amount of added water was thirty centimeters depth of water. Such water was divided into two equal doses 15 cm each and allowed to pass through the soil column. Termination of leaching was done after passing the two equal doses. The leachates were collected and analyzed and soil sample were taken from each column at three layers 0 - 20, 20 - 40 and 40 - 60 cm depth using cylindroids tube of 2 cm inside diameter. Each sample was air dried, crushed, mixed and passed through a 2 mm sieves and analyzed for salinity, soluble ions, pH and exchangeable cations. The 2 cm hole was filled with a mixture of sand and hot wax. A halophyte plant (*Atriplex halimus*) was planted in each column after termination of leaching and received all necessary practices. At the end of six month growing

period plants were harvested and some plant morphological and chemical characters were recorded. Then soil columns were separated into 3 segments 0-20, 20-40 and 40-60 cm. Soil of each segment was air dried, crushed, mixed and passed through a 2 mm sieves and analyzed for salinity, soluble ions, pH and exchangeable cations.

Each of the determined parameters either in leachates or soil was passed through the tetra factorial computer programmed model of Moussa and Youssef (1992) in which the results of all the possible combination were printed either on the surface or inside the tetrahedron; each value refers to specific combination according to its position on or inside the tetrahedron. The total numbers of the output values are 165. The values located on the surface area of the principle tetrahedron amount to 130 corresponding to 4 single, 21 double and 105 triple-factor treatments. The other 35 located inside the principle one and refer to quadruple-factor. It is worthy to mention that any of the 165 intersections have the same cost of the amendments, which considered highly important when comparing the effect and inter effect of the materials from the economical

Table 1. Physical and chemical properties of the studied soil and chemical properties of water used for irrigation (El-Salam canal water)

Property	Soil	Water
<b>Particle size distribution [%]</b>		
▣ Clay	38.54	
▣ Silt	10.14	
▣ Fine sand	35.80	
▣ Coarse sand	15.52	
▣ Texture class [According to USDA triangle]	Sandy Clay	
<b>Soil moisture characteristics [%]</b>		
▣ Saturation percent	33.76	
▣ Field capacity	16.89	
▣ Wilting point	8.44	
<b>density (Mg.m<sup>-1</sup>)</b>		
▣ Bulk density	1.65	
▣ Particle density	2.56	
<b>Organic matter [g kg<sup>-1</sup>]</b>	5.30	
<b>CaSO<sub>4</sub> [g kg<sup>-1</sup>]</b>	15.00	
<b>CaCO<sub>3</sub> [g kg<sup>-1</sup>]</b>	41.40	
<b>Soluble ions, EC and pH</b>		
▣ EC (dSm <sup>-1</sup> ) [Soil extract 1:2.5]	28.80	1.72
▣ Soil reaction (pH) [Soil suspension 1:2.5]	8.40	7.60
▣ Soluble ions (mmol <sub>c</sub> L <sup>-1</sup> ) [Soil extract 1:2.5]		
▪ Na <sup>+</sup>	135.72	10.47
▪ K <sup>+</sup>	6.09	0.25
▪ Ca <sup>++</sup>	96.00	3.69
▪ Mg <sup>++</sup>	123.20	2.79
▪ Cl <sup>-</sup>	259.54	13.62
▪ CO <sub>3</sub> <sup>=</sup>	00.00	00.00
▪ HCO <sub>3</sub> <sup>-</sup>	2.24	1.06
▪ SO <sub>4</sub> <sup>=</sup>	99.23	2.52
▪ SAR	12.96	5.82
<b>Exchangeable cations and CEC (cmol<sub>c</sub> kg<sup>-1</sup>)</b>		
▣ Na <sup>+</sup>	10.54	
▣ K <sup>+</sup>	2.83	
▣ Ca <sup>++</sup>	5.89	
▣ Mg <sup>++</sup>	8.46	
▣ CEC (cmol <sub>c</sub> kg <sup>-1</sup> )	27.75	
<b>Exchangeable sodium percentage (ESP)</b>	38.00	

Note: Converting EC dSm<sup>-1</sup> to mmol<sub>c</sub> L<sup>-1</sup> soluble salts (or cations) agrees with the equation of Gupta (1990), getting a factor of 12 to 13 for EC > 5; also Tanji (1990) gives an equation predicting mmol<sub>c</sub>L<sup>-1</sup> from EC dSm<sup>-1</sup> for EC > 5.

Table 2. Properties of amendments used for reclamation

Property	Value
1- Sulphuric acid	
▪ Concentration [%]	95
▪ Specific gravity	1.834
2- Gypsum [CaSO <sub>4</sub> .2H <sub>2</sub> O]	
▪ Purity [%]	98.81
▪ pH [1:5 water]	7.80
▪ EC [1:5 water]	2.56
▪ Ca [g kg <sup>-1</sup> ]	229.8
▪ S [g kg <sup>-1</sup> ]	175.0
3- Limestone [CaCO <sub>3</sub> ]	
▪ Purity [%]	97.50
▪ Moisture [%]	0.20
▪ pH	8.10
▪ Ca [g kg <sup>-1</sup> ]	390.0
▪ CaO [g kg <sup>-1</sup> ]	550.0
4- Compost [made from plant]	
▪ Moisture [%]	23.00
▪ pH [1:10]	7.00
▪ EC dS/m [1:10]	2.56
▪ Total N [g kg <sup>-1</sup> ]	21.50
▪ Total P [g kg <sup>-1</sup> ]	10.90
▪ Total K [g kg <sup>-1</sup> ]	3.10
▪ Organic mater [g kg <sup>-1</sup> ]	672.0
▪ C/N Ratio	1:18

Table 3. Treatment codes and ratios of amendments along with their rates in each treatment Feddan<sup>-1</sup> in the current study

Treat. No.	Amendments codes and ratios				Amendments codes and rates [Mg.Fed-1]			
	X1	X2	X3	X4	X1	X2	X3	X4
	[G]	[L]	[S]	[C]	[G]	[L]	[S]	[C]
1	8	0	0	0	13.50	00.00	00.00	00.00
2	0	8	0	0	00.00	5.80	00.00	00.00
3	0	0	8	0	00.00	00.00	0.25	00.00
4	0	0	0	8	00.00	00.00	00.00	8.12
5	2	2	2	2	3.38	1.45	0.06	2.03
6	4	4	0	0	6.75	2.90	00.00	00.00
7	4	0	4	0	6.75	00.00	0.13	00.00
8	4	0	0	4	6.75	00.00	00.00	4.06
9	0	4	4	0	00.00	2.90	0.13	00.00
10	0	4	0	4	00.00	2.90	00.00	4.06
11	0	0	4	4	00.00	00.00	0.13	4.06
12	5	1	1	1	8.44	0.73	0.03	1.02
13	1	5	1	1	1.69	3.63	0.03	1.02
14	1	1	5	1	1.69	0.73	0.16	1.02
15	1	1	1	5	1.69	0.73	0.03	5.06
16	2.5	2.5	2.5	0.5	4.22	1.81	0.08	0.51
17	2.5	2.5	0.5	2.5	4.22	1.81	0.02	2.54
18	2.5	0.5	2.5	2.5	4.22	0.36	0.08	2.54
19	0.5	2.5	2.5	2.5	0.84	1.81	0.08	2.54

Notes: X1: Gypsum "G", X2: Lime-stone "L", X3: Sulfuric Acid "S" and X4: Compost "C"

point of view. Moreover the average value, general mean error, correlation coefficient, fisher, criterion, optimum combination and adequacy test of the model through the treatments 16-19 are of the programmed output. Physical and chemical analyses were done according to Baruah and barthakur (1997), Page (1991) and Chapman and Pratt (1961).

## RESULTS AND DISCUSSION

### Effect of the Leaching

Most of the reclamation indicators at the end of both, leaching process and harvesting of halophyte plant as affected by all the possible combination of gypsum "G", lime "L", sulfuric acid "S" and compost "C" were significantly decreased comparing with the original soil (Table 4). However, the response took the same trend.

The effect of EC  $\text{dSm}^{-1}$ , SAR, and ESP soil properties have the same trend. There was a high prohibitive significant correlation between EC and each of SAR and ESP  $r = 0.99$  and  $0.98$ , respectively. Soil electrical conductivity (EC,  $\text{dSm}^{-1}$ ) was chosen to represent this group of soil parameters.

### Soil salinity and sodicity

Data of Table 4 and Fig. 1 show that the EC ranged between  $7.07$  and  $16.77 \text{ dSm}^{-1}$  with a general mean  $10.95 \text{ dSm}^{-1}$ . However the located values of the four single treatments were  $15.67$ ,  $16.77$ ,  $16.43$  and  $14.93 \text{ dSm}^{-1}$  for G, L, S and C, respectively. These results suggest that C gave the promotion effect compared with the other amendments, the order where  $C > G > S > L$ . These results could explain that C has pronounced role in decreasing soil salinity.

Scanning the different values of Fig. 1A shows  $8.6 \text{ dSm}^{-1}$  as the minimum value, corresponding to an interpolated four combined treatments consisting of [0: 2: 3: 3] (of the 8 points score) of [G, L, S & C], respectively. This result indicates the significant effect of G, L, S and C at the rate [0.00, 1.45, 0.09 & 3.05]  $\text{Mgfd}^{-1}$ , respectively. Scanning the different values of Fig. 1B shows that the value  $5.80 \text{ dSm}^{-1}$  is the minimum one, corresponding to an interpolated four combined treatments consisting of [1.0: 2.0: 3.0: 2.0] (of the 8 points score), respectively of [G, L, S and C], respectively. This result reflects the marked effect of the combination of G, L, S and C at the rate [1.69, 1.45, 0.09 & 2.03]  $\text{Mgfd}^{-1}$ . Fig. 1C indicate that the

center point of the tetrahedron has an EC value  $14.10 \text{ dSm}^{-1}$  corresponding to treatment of [2: 2: 2: 2] i.e. equivalent mixture of the four amendments. The original soil was saline sodic and very compact. Addition of organic must have loosened the soil and increased its porosity (Ahmed *et al.*, 1988). Addition of gypsum and/or FYM must have enhanced the chemical reaction and exchanged the sodium ions with  $\text{Ca}^{2+}$  on the soil exchange complex. The  $\text{Na}^+$  ions in soluble form would leach down due to improved soil physical conditions.

Generally, Fig. 1 shows that the optimum combination for decreasing EC was obtained by the mixture of G, L, S and C at the ratio of [1.0: 2.0: 3.0: 2.0] (of the 8 points score) [G, L, S and C], respectively. This result reflects significant effect of G, L, S and C at the rate [1.69, 1.45, 0.09 and 2.03]  $\text{Mg fed}^{-1}$ , respectively. The pattern of EC was similar to each of the soluble cations and soluble anions where data reveal highly significant correlation between EC and each of the cations and anions. Niazi *et al.*, (2001), confirmed in field study that gypsum (100% GR) added with FYM and gypsum 75% and  $\text{H}_2\text{SO}_4$  increased the yield for the first year of rice crop.

Regarding the effect on SAR, the optimum combination for decreasing it, was obtained by a mixture of G, L, S and C at the ratio of [1.0: 2.0: 3.0: 2.0] (of the 8 points score), respectively. This result reflects the marked effect of G, L, S and C at the rate [1.69, 1.45, 0.09 & 2.03]  $\text{Mg fed}^{-1}$ , respectively. Regarding ESP the optimum combination for its decrease was obtained by a mixture of G, L, S and C at the ratio of [2.0: 1.0: 3.0: 2.0] (of the 8 points score) of [G, L, S and C], respectively. This reflects the marked effect of G, L, S and C at the rate [3.38, 0.73, 0.09 & 2.03]  $\text{Mg fed}^{-1}$ , respectively. Chaganti, (2008) examined the effect of "gypsum, sulfur, and poultry manure" on EC, SAR and ESP in soils and found significant decreases in soil SAR and ESP in the 0 - 5 cm depth.

#### Soil reaction (soil pH)

Regarding soil pH (Table 4 and Fig. 2). Results (Fig. 2A) show that a pH value of 7.99 was the minimum one, corresponding to an interpolated four combined treatments consisting of [0: 0: 0: 8] (of the 8 points score) of [G, L, S and C], respectively. This reflects the positive effect of G, L, S and C at the rate [0.00, 0.00, 0.00 & 8.12]  $\text{Mg fed}^{-1}$ , respectively in decreasing



soil pH. Fig. 2B shows a pH value of 8.02 as the minimum one, corresponding to an interpolated treatments consisting of [5: 1: 1: 1] (of the 8 points score) of [G, L, S and C], respectively. This is the effect of G, L, S and C at the rate [8.44, 0.73, 0.03 & 1.02] Mg $\text{fed}^{-1}$ , respectively.

Fig. 2C indicates that the center point of the tetrahedron has a pH value 8.12 corresponding to treatment [2: 2: 2: 2] of the four amendments.

Fig. 2 shows the optimum combination for decreasing pH was obtained by a mixture of G, L, S and C at the ratio of [0: 0: 0: 8] (of the 8 points score) respectively of [G, L, S and C], respectively. This shows the significant effect of G, L, S and C at the rate [0.00, 0.00, 0.00 & 8.12] Mg $\text{fed}^{-1}$ , respectively. Avinelech *et al.*, (1990) concluded that applying organic composts to saline sodic soils would help in chelating calcium and reduced pH leading to an increase in solubility of  $\text{CaCO}_3$  and preventing  $\text{Ca}^{++}$  precipitation. Anand (1992) stated that organic amendments decreased soil sodicity and increased  $\text{pCO}_2$  and exchangeable  $\text{Ca}^{++}$  and  $\text{Mg}^{++}$ . Benz *et al.*, (1987) reported that straw mulches effectively used for saline sodic reclamation.

### Effect after Growth

Regarding to soil parameters after harvesting of the halophyte plant Table 4 shows considerable positive effects of amendments compared with the effect after leaching; data after plant harvest shows more improvement of soil properties. Leaching and halophyte growing decreased soil salinity by about two thirds. These results are in harmony with those of Al-Khateeb *et al.*, (2001).

### Salinity and SAR

Values of EC and SAR have the same trend there was a high correlation between SAR and EC  $r = 0.99$ . Hence, the different soil reclamation characters (i.e. pH, ESP and bulk density) can be illustrated by one of them. Soil electrical conductivity (EC,  $\text{dSm}^{-1}$ ) was chosen to represent this group of soil parameters.

Data of Table 4 and Fig. 3 indicate that EC ranged between 13.14 and 3.60  $\text{dSm}^{-1}$  with a general mean 7.27  $\text{dSm}^{-1}$ . Values for the four single treatments were 12.07, 13.14, 12.75 and 11.18  $\text{dSm}^{-1}$  for of G, L, S and C, respectively. Thus C gave the most effect compared with the others, the order being  $C > G > S > L$ .

Scanning the different values, Fig 3A shows that 4.80  $\text{dSm}^{-1}$  this minimum one, regarding the

interpolated combined treatment of [0: 2: 3: 3] (of the 8 points score) of [G, L, S & C], respectively. This is a treatment of G, L, S and C at the rate [0.00, 1.45, 0.09 & 3.05] Mg $\text{fed}^{-1}$ , respectively. Fig. 3B shows 2.10 dSm $^{-1}$  as the minimum one, regarding the interpolated treatment of [1.0: 2.0: 3.0: 2.0] (of the 8 points score), respectively of [G, L, S and C], respectively. Thus G, L, S and C at the rate [1.69, 1.45, 0.09 & 2.03] Mg $\text{fed}^{-1}$ , was very effective. Fig. 3C indicates that the center point of the tetrahedron has an EC 5.84 dSm $^{-1}$  corresponding to the treatment of [2: 2: 2: 2]. The pattern of each of the soluble cations [i.e. Na $^{+}$ , K $^{+}$ , Ca $^{++}$  and Mg $^{++}$ ] and soluble anions [i.e. Cl $^{-}$ , HCO $_3^{-}$  and SO $_4^{-}$ ] was similar to that of EC.

Fig. 3 shows the optimum combination for decreasing EC as well as SAR is a mixture of G, L, S and C at the rate of [1.0: 2.0: 3.0: 2.0] (of the 8 points score), respectively. This is treatment of [1.69, 1.45, 0.09 and 2.03] Mg/fed, respectively.

#### Soil pH and ESP

Regarding to soil pH after harvesting data of Table 4 and Fig. 4 show that soil pH was decreased by the different combinations of G, L, S and C. Fig. 4A shows a pH

value 7.70 as regarding an interpolated four combined treatments consisting of [0: 0: 8: 0] (of the 8 points score) of [G, L, S and C], respectively, while Fig. 4B shows a value of 7.84 as the minimum one. Fig. 4C shows that the center point of the tetrahedron has a pH 7.92 for the treatment of [2: 2: 2: 2]. Fig. 4 shows the optimum combination for decreasing pH was obtained by a mixture of G, L, S and C at the ratio of [0: 0: 8: 0] (of the 8 points score) of [G, L, S and C], respectively. This shows a marked effect of applying sulfuric acid at a rate of 0.25 Mg $\text{fed}^{-1}$ . The general mean of pH decreased from 8.40 to 7.99 to 7.70 at the end of the leaching process followed by halophyte planting.

Regarding ESP after harvesting, data of Table 4 and Fig. 5 show that ESP was affected by application of amendments. Scanning the different values of Fig. 5A show that ESP 8.80 as the minimum one as interpolated for the treatment of [5: 3: 0: 0] (of the 8 points score) of [G, L, S and C], respectively, while Fig. 5B shows that ESP 9.10 as the minimum one. Fig. 5C indicates that the center point of the tetrahedron has an ESP of 9.70 corresponding to treatment of [2: 2: 2: 2]. Fig. 5

Table 4. Some soil chemical and physical indicators at the end of both, leaching process and harvesting of halophyte plant as affected by different combination gypsum, lime stone, sulfuric acid and organic compost

Treatments				At the end of leaching process				After harvesting						
X1	X2	X3	X4	EC dSm <sup>-1</sup>	pH	SAR	ESP	BC dSm <sup>-1</sup>	pH	SAR	ESP	BD	% P	
8	0	0	0	15.67	8.07	9.56	20.43	12.07	8.01	8.39	9.44	1.29	51.20	
0	8	0	0	16.77	8.02	9.89	22.16	13.14	7.84	8.76	10.85	1.30	51.07	
0	0	8	0	16.43	8.01	9.79	21.33	12.75	7.70	8.63	9.97	1.27	51.90	
0	0	0	8	14.93	7.99	9.33	19.75	11.18	7.85	8.08	9.92	1.26	52.46	
2	2	2	2	14.13	8.12	9.08	19.22	10.48	7.92	7.79	9.67	1.29	51.27	
4	4	0	0	13.47	8.14	8.86	18.86	9.69	7.96	7.52	8.89	1.25	52.83	
4	0	4	0	12.53	8.26	8.55	17.86	8.93	8.01	7.22	8.87	1.28	51.81	
4	0	0	4	11.07	8.29	8.03	17.36	7.29	7.89	6.52	9.80	1.27	52.04	
0	4	4	0	11.00	8.30	8.01	17.67	7.15	7.97	6.46	11.44	1.30	50.93	
0	4	0	4	10.53	8.33	7.84	16.94	6.94	8.04	6.36	11.08	1.27	52.05	
0	0	4	4	9.93	8.12	7.61	16.52	6.12	7.98	5.97	9.99	1.28	51.76	
5	1	1	1	7.43	8.02	6.58	13.49	3.64	7.83	4.60	9.12	1.25	52.66	
1	5	1	1	8.33	8.10	6.97	15.81	4.76	7.91	5.27	11.21	1.25	52.66	
1	1	5	1	7.07	8.07	6.42	13.08	3.37	7.95	4.44	10.04	1.30	50.76	
1	1	1	5	7.73	8.07	6.72	14.01	4.14	7.88	4.91	9.81	1.26	52.55	
2.5	2.5	2.5	0.5	7.27	8.08	6.51	14.22	3.60	7.98	4.58	8.86	1.26	52.27	
2.5	2.5	0.5	2.5	8.83	8.11	7.18	16.02	5.31	7.85	5.56	9.27	1.25	52.88	
2.5	0.5	2.5	2.5	7.22	8.12	6.49	12.46	3.33	7.90	4.41	9.42	1.26	52.62	
0.5	2.5	2.5	2.5	7.73	8.12	6.72	15.60	4.24	7.97	4.97	9.43	1.27	52.09	
Correlation coefficient with EC					-0.11	0.99	0.97							
Correlation coefficient with BC									0.23	0.99	0.12			
Correlation coefficient with BD														-0.98

Notes: X1: Gypsum, X2: Lime-stone, X3: Sulfuric Acid and X4: Compost

Table 5. Optimal combination for the different soil character (of the 8 point score and Mg fed<sup>-1</sup>) from gypsum, limestone, sulfuric acid and compost

Soil property	Initial value	Optimum combination [After leaching process]			Optimum combination [After harvesting]			
		Ratio	Mg fed <sup>-1</sup>	Value	Ratio	Mg fed <sup>-1</sup>	Value	
		[X1:X2:X3:X4]	[X1:X2:X3:X4]		[X1:X2:X3:X4]	[X1:X2:X3:X4]		
EC [dSm <sup>-1</sup> ]	28.80	[1:2:3:2]	1.69:1.45:0.09:2.03	5.80	[1:2:3:2]	1.69:1.45:0.09:2.03	2.10	
pH	8.40	[0:0:0:8]	0.00:0.00:0.00:8.12	7.99	[0:0:8:0]	0.00:0.00:0.25:0.00	7.70	
SAR	12.99	[1:1:4:2]	1.69:0.73:0.13:2.03	6.05	[1:2:3:2]	1.69:1.45:0.09:2.03	4.06	
ESP	38.01	[2:1:3:2]	3.38:0.73:0.09:2.03	12.4	[5:3:0:0]	8.44:2.18:0.00:0.00	8.80	
B.D [Mgm <sup>-1</sup> ]	1.65	Not Determined				[3:3:1:1]	5.06:2.18:0.03:1.02	1.24
Recommended combination		[1:1:2.5:3.5]	1.70:0.73:0.08:3.60		[2:2:3:1]	3.38:1.45:0.09:1.02		

Notes: B.D: Bulk Density, X1: Gypsum, X2: Limestone, X3: Sulfuric Acid and X4: Compost.

NUMBER OF TREATMENTS = 19; NUMBER OF REPLICATES = 3

## EXPERIMENTAL DATA AND (MEAN)

1.15.66	15.62	15.72	(15.67)	2.16.84	16.87	16.59	(16.77)
3.16.41	16.36	16.53	(16.43)	4.14.95	14.94	14.80	(14.93)
5.14.08	14.12	14.19	(14.13)	6.13.55	13.44	13.41	(13.47)
7.12.52	12.41	12.66	(12.53)	8.11.10	11.08	11.03	(11.07)
9.11.07	11.00	10.93	(11.00)	10.10.51	10.51	10.58	(10.53)
11.9.91	9.97	9.92	(9.93)	12.7.49	7.52	7.29	(7.43)
13.8.38	8.59	8.03	(8.33)	14.7.08	7.08	7.04	(7.07)
15.7.66	7.74	7.79	(7.73)	16.7.27	7.30	7.23	(7.27)
17.8.81	8.94	8.75	(8.83)	18.7.35	7.23	7.88	(7.22)
19.7.74	7.69	7.77	(7.73)				

GENERAL MEAN = 10.9514 MEAN ERROR = 4.820766E<sup>-41</sup>

CORRELATION COEFFICIENT = 0.9997072

KERETRION FISHER F(18 38) = 3332.657

## REALITY ADEQUATE TEST OF THE MODEL THROUGH THE LAST FOUR TREATMENTS

EXPERIMENTAL TREAT.	TABULAR	DIFFERENCE	KER. STUDENT
16	7.27 - 7.08 =	0.18	3.18
17	8.83 - 6.96 =	1.88	32.47
18	7.22 - 6.39 =	0.92	15.98
19	7.73 - 5.87 =	1.87	32.27

## THE SIGNIFICANT ADEQUATE SECTOR OF THE SQUARE TEST FOR THE CALCULATED VALUE, MULTIPLY BY 10

TETRAHYDRON WITH X<sub>1</sub> minimum = 0; X<sub>1</sub> maximum = 1

X4	X1	X4
149	132 119 112 111 114 123 137 157 137 123 134 111 112 119 132 149	X4
	128 115 107 104 107 115 128 146 142 124 111 104 101 104 112 126	
	114 105 101 103 109 121 139 132 132 115 104 97 96 100 110	
	106 102 102 108 119 135 125 122 127 111 100 95 96 101	
	105 105 109 119 135 122 116 117 125 111 102 98 99	
	111 115 124 138 122 113 113 116 128 115 107 105	
	123 131 144 125 114 109 111 120 136 124 117	
	142 154 132 117 110 109 115 128 146 137	
X2	168 143 125 114 110 113 123 140 164	X3 Fig. (1A)
	142 120 106 98 97 104 117 137	
	123 105 93 89 91 101 117	
	111 96 87 92 105	
	105 93 88 90 99	
	106 98 96 101	
	114 108 110	
	128 126	
	149	
	X4	

TETRAHYDRON WITH X<sub>1</sub> minimum = .125 X<sub>1</sub> maximum = .625

X4	X1	X4
77	69 65 67 74 67 65 69 77	X4
	69 64 65 71 67 61 60 65	
	67 67 72 64 64 59 60	
	72 76 65 62 65 62	
X2	83 70 63 63 71	X3 Fig. (1B)
	72 61 62	
	67 60 60	
	69 65	
	77	
	X4	

TETRAHYDRON WITH X<sub>1</sub> minimum = .25; X<sub>1</sub> maximum = .25

X4	X1	X4
	X4 X4	
	X2 X3	Fig. (1C)
	X4	

MINIMUM VALUE YR = 5.80  
OPTIMUM COMBINATION: (1.0; 2.0; 3.0; 2.0)Fig. 1. Computer output of EC [dSm<sup>-1</sup>] at the end of leaching process as affected by different combination gypsum, limestone, sulfuric acid and organic compost

NUMBER OF TREATMENTS = 19; NUMBER OF REPLICATES = 3

## EXPERIMENTAL DATA AND (MEAN)

1.8.02	8.06	8.12	(8.07)	2.7.97	8.00	8.09	(8.02)
3.7.97	8.00	8.07	(8.01)	4.7.96	7.94	8.04	(7.99)
5.8.12	8.13	8.12	(8.12)	6.8.16	8.16	8.11	(8.14)
7.8.22	8.25	8.32	(8.26)	8.8.24	8.27	8.36	(8.29)
9.8.26	8.28	8.37	(8.30)	10.8.29	8.38	8.39	(8.33)
11.8.86	8.15	8.15	(8.12)	12.7.97	8.00	8.09	(8.02)
13.8.80	8.10	8.19	(8.10)	14.8.04	8.05	8.11	(8.07)
15.8.02	8.09	8.10	(8.07)	16.8.02	8.07	8.14	(8.08)
17.8.07	8.10	8.16	(8.11)	18.8.09	8.14	8.12	(8.12)
19.8.08	8.12	8.17	(8.12)				

GENERAL MEAN = 8.122982 MEAN ERROR = 2.476396E<sup>-33</sup>

CORRELATION COEFFICIENT = 0.9175124

KERETRION FISHER F(18 38) = 11.2359

## REALITY ADEQUATE TEST OF THE MODEL THROUGH THE LAST FOUR TREATMENTS

EXPERIMENTAL TREAT.	TABULAR	DIFFERENCE	KER. STUDENT
16	8.08 - 8.20 =	-0.12	-4.00
17	8.11 - 8.21 =	-0.10	-3.44
18	8.12 - 8.18 =	-0.07	-2.26
19	8.12 - 8.23 =	-0.10	-3.52

## THE SIGNIFICANT ADEQUATE SECTOR OF THE SQUARE TEST FOR THE CALCULATED VALUE, MULTIPLY BY 100

TETRAHYDRON WITH X<sub>1</sub> minimum = 0; X<sub>1</sub> maximum = 1

X4	X1	X4
809	811 820 826 829 828 824 817 807 817 824 828 829 826 820 811	X4
	813 823 829 832 831 828 821 810 816 824 829 831 829 824 816 804	
	824 830 833 833 830 823 813 819 822 828 831 830 826 819 808	
	830 834 834 831 824 814 822 825 826 829 830 827 820 811	
	833 833 830 824 814 823 827 829 826 828 826 821 812	
	831 828 822 813 823 828 831 829 824 823 819 812	
	825 820 811 821 828 831 831 827 819 816 810	
	816 807 818 826 831 831 828 822 812 806	
X2	802 814 823 829 830 828 823 814 881	X3 Fig. (2A)
	816 825 831 833 832 827 818 806	
	825 832 835 834 829 821 810	
	831 834 834 830 823 812	
	833 833 830 823 812	
	830 827 821 811	
	824 818 808	
	813 804	
	X4	

TETRAHYDRON WITH X<sub>1</sub> minimum = .125 X<sub>1</sub> maximum = .625

X4	X1	X4
807	811 811 808 808 811 811 807	X4
	814 814 812 806 807 811 812 809	
	816 814 808 811 810 812 810	
	815 810 813 813 810 809	
X2	810 814 815 813 807	X3 Fig. (2B)
	815 817 815 809	
	816 815 810	
	814 809	
	807	
	X4	

TETRAHYDRON WITH X<sub>1</sub> minimum = .25; X<sub>1</sub> maximum = .25

X4	X1	X4
	X4 X4	
	X2 X3	Fig. (2C)
	X4	

MINIMUM VALUE YR = 7.99  
OPTIMUM COMBINATION: (0; 0; 0; 8)

Fig. 2. Computer output of pH at the end of leaching process as affected by different combination gypsum, limestone, sulfuric acid and organic compost

NUMBER OF TREATMENTS = 19; NUMBER OF REPLICATES = 3  
EXPERIMENTAL DATA AND (MEAN)

1.12,11	12.11	11.99	(12.07)	2.13,18	13.15	13.10	(13.14)
3.12,77	12.77	12.72	(12.75)	4.11,17	11.25	11.13	(11.18)
5.10,49	10.34	10.38	(10.40)	6.9,73	9.73	9.61	(9.69)
7.8,93	8.94	8.93	(8.93)	8.7,35	7.34	7.18	(7.29)
9.7,16	7.15	7.14	(7.15)	10.6,95	6.95	6.93	(6.94)
11.6,15	6.14	6.06	(6.12)	12.3,63	3.65	3.63	(3.64)
13.4,76	4.75	4.76	(4.76)	14.3,37	3.39	3.37	(3.38)
15.4,17	4.17	4.08	(4.14)	16.3,64	3.66	3.52	(3.61)
17.5,38	5.29	5.15	(5.31)	18.3,36	3.41	3.22	(3.33)
19.4,30	4.15	4.25	(4.23)				

GENERAL MEAN = 7.266492 MEAN ERROR = 2.987955E<sup>-01</sup>  
CORRELATION COEFFICIENT = 0.9998784  
KERETRIION FISHER F(18 38) = 8685.298

REALITY ADEQUATE TEST OF THE MODEL THROUGH THE LAST FOUR TREATMENTS

EXPERIMENTAL TREAT.	TABULAR	DIFFERENCE	KER-STUDENT
16	3.61 - 3.34 =	0.27	7.45
17	5.31 - 3.26 =	2.04	56.96
18	3.33 - 2.57 =	0.76	21.14
19	4.23 - 2.19 =	2.05	57.12

THE SIGNIFICANT ADEQUATE SECTOR OF THE SQUARE TEST FOR THE CALCULATED VALUE, MULTIPLY BY 10

TETRAHYDRON WITH X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

X-minimum = 0; X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub> X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

112 94 82 75 73 77 86 101 121 101 86 77 73 75 82 94 112 X4

91 78 69 67 69 77 90 109 106 87 74 66 63 66 75 88

78 68 64 65 72 84 102 95 96 79 66 60 58 62 72

70 65 65 70 81 97 88 86 91 74 63 58 58 63

69 68 72 82 97 84 79 80 89 74 64 60 61

75 78 86 100 84 75 74 79 92 78 70 67

87 94 107 88 76 71 73 83 100 87 80

106 117 95 79 71 70 77 91 111 100

X2 131 106 87 76 72 75 85 103 128 X3 Fig. (3A)

106 84 68 60 59 65 79 100

87 68 56 51 53 63 80

75 59 50 54 67

69 56 51 52 61

70 61 58 63

78 71 72

91 88

112

X4

TETRAHYDRON WITH X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

X-minimum = .125 X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub> X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub> X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

41 32 28 29 36 29 28 32 41 X4

33 28 28 34 29 24 23 19

31 30 35 27 26 22 23

36 39 28 24 28 25

X2 48 33 26 26 34 X3 Fig. (3B)

36 25 25

31 24 23

33 29

41

X4

TETRAHYDRON WITH X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

X-minimum = .25; X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub> X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub> X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

54 X4 X4 Fig. (3C)

54 X4

X2 X3

X4

MINIMUM VALUE YR = 2.10  
OPTIMUM COMBINATION: (1; 2; 3; 2)

Fig. 3. Computer output of EC [dSm<sup>-1</sup>] after harvesting of halophyte plant as affected by different combination gypsum, limestone, sulfuric acid and organic compost

NUMBER OF TREATMENTS = 19; NUMBER OF REPLICATES = 3  
EXPERIMENTAL DATA AND (MEAN)

1.7,99	8.00	8.03	(8.01)	2.7,80	7.88	7.85	(7.84)
3.7,71	7.61	7.77	(7.70)	4.7,80	7.85	7.90	(7.85)
5.7,92	7.90	7.95	(7.92)	6.7,98	7.96	7.92	(7.95)
7.8,00	8.05	7.98	(8.01)	8.7,97	7.93	7.77	(7.89)
9.7,92	8.00	7.98	(7.97)	10.8,06	8.07	8.80	(8.04)
11.7,98	7.96	8.00	(7.98)	12.7,76	7.87	7.87	(7.83)
13.7,84	7.98	7.92	(7.91)	14.7,91	7.96	7.97	(7.95)
15.7,76	7.93	7.95	(7.86)	16.7,99	7.96	8.00	(7.98)
17.7,85	7.78	7.93	(7.85)	18.7,88	7.92	7.91	(7.90)
19.7,97	7.98	7.95	(7.97)				

GENERAL MEAN = 7.91907 MEAN ERROR = 2.567601E<sup>-01</sup>  
CORRELATION COEFFICIENT = 0.8726434  
KERETRIION FISHER F(18 38) = 6.740749

REALITY ADEQUATE TEST OF THE MODEL THROUGH THE LAST FOUR TREATMENTS

EXPERIMENTAL TREAT.	TABULAR	DIFFERENCE	KER-STUDENT
16	7.98 - 7.97 =	-0.01	0.37
17	7.85 - 7.93 =	-0.07	-2.34
18	7.90 - 7.95 =	-0.05	-1.53
19	7.97 - 8.02 =	-0.05	-1.68

THE SIGNIFICANT ADEQUATE SECTOR OF THE SQUARE TEST FOR THE CALCULATED VALUE, MULTIPLY BY 100

TETRAHYDRON WITH X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

X-minimum = 0; X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub> X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

785 785 786 787 789 791 794 797 801 797 794 791 789 787 786 785 785 X4

794 793 793 793 794 796 798 800 804 801 798 796 794 793 792 792

800 798 797 797 797 798 799 803 805 802 800 799 797 797 797

803 801 799 798 797 797 802 804 804 802 800 799 799 799

804 801 799 797 795 800 803 803 801 800 799 798 798

803 799 796 793 799 802 802 800 796 795 795 795

799 795 791 796 800 801 800 796 789 789 789

793 788 794 798 799 798 795 789 780 781

X2 784 791 795 797 797 794 788 780 X3 Fig. (4A)

793 799 802 802 801 796 790 781

799 804 806 805 802 797 789

803 806 807 805 801 795

804 806 806 803 798

803 804 803 799

800 799 797

794 792

785

X4

TETRAHYDRON WITH X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

X-minimum = .125 X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub> X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub> X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

788 786 785 783 783 785 786 788 X4

793 790 788 786 789 790 792 794

795 791 788 792 793 794 797

794 790 794 796 795 797

X2 791 796 798 798 795 X3 Fig. (4B)

794 798 798 797

795 797 797

793 794

788

X4

TETRAHYDRON WITH X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

X-minimum = .25; X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub> X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub> X<sub>1</sub> X<sub>2</sub> X<sub>3</sub> X<sub>4</sub>

804 X4 X4 Fig. (4C)

804 X4

X2 X3

X4

MINIMUM VALUE YR = 7.7  
OPTIMUM COMBINATION: (0; 0; 8; 0)

Fig. 4. Computer output of pH after harvesting of halophyte plant as affected by different combination gypsum, limestone, sulfuric acid and organic compost

NUMBER OF TREATMENTS = 19; NUMBER OF REPLICATES = 3

EXPERIMENTAL DATA AND (MEAN)

1. 8.45	9.66	10.22	(9.44)	2. 11.47	9.82	11.27	(10.85)
3. 10.29	10.41	9.20	(9.97)	4. 11.68	9.24	8.85	(9.92)
5. 9.67	10.14	9.21	(9.67)	6. 9.49	9.46	7.72	(8.89)
7. 9.04	9.46	8.09	(8.86)	8. 8.80	10.50	10.10	(9.80)
9. 11.54	10.37	12.41	(11.44)	10. 11.78	10.87	10.58	(11.06)
11. 10.99	8.77	10.22	(9.99)	12. 9.21	9.62	9.13	(9.12)
13. 11.72	11.28	10.62	(11.21)	14. 9.49	10.47	10.17	(10.04)
15. 9.50	10.37	9.56	(9.81)	16. 8.85	8.85	8.89	(8.86)
17. 9.51	9.79	8.51	(9.27)	18. 8.36	10.11	9.80	(9.42)
19. 8.96	10.14	9.19	(9.43)				

GENERAL MEAN = 9.846842 MEAN ERROR = 0.3774801

CORRELATION COEFFICIENT = 0.7621138

KRUTRIKON FISHER F(18,38) = 2.925146

REALITY ADEQUATE TEST OF THE MODEL THROUGH THE LAST FOUR TREATMENTS

	EXPERIMENTAL TREATMENT - TABULAR	DIFFERENCE	KER-STUDENT
16	8.86 - 9.90 =	-1.04	-2.29
17	9.27 - 9.99 =	-0.72	-1.59
18	9.42 - 9.53 =	-0.11	-0.24
19	9.43 - 10.84 =	-1.41	-3.11

THE SIGNIFICANT ADEQUATE SECTOR OF THE SQUARE TEST FOR THE CALCULATED VALUE, MULTIPLY BY 100

TETRAHYDRON WITH		X-minimum = 0;	X-maximum = 1													
		X1														
X4			X4													
990	990	990	990	980	970	970	960	940	960	970	970	980	990	990	990	990
1030	1020	1010	990	970	950	930	910	910	930	940	960	970	980	990	990	990
1070	1040	1010	980	950	920	890	900	890	920	930	950	970	980	1000		
1090	1050	1010	970	920	890	890	900	890	910	930	960	980	1000			
1110	1060	1000	950	890	910	910	910	890	920	950	970	1000				
1110	1050	980	910	940	950	940	930	900	930	970	1000					
1110	1040	960	980	1000	1000	980	960	920	960	1000						
1100	1010	1040	1060	1060	1060	1030	1000	950	1000							
X2	1090	1120	1140	1150	1140	1130	1100	1050	1000	X3	Fig. (5A)					
	1100	1130	1140	1140	1120	1090	1050	1000								
	1110	1130	1130	1110	1090	1050	1000									
	1120	1120	1110	1080	1050	1000										
	1110	1100	1080	1050	1000											
	1090	1070	1040	1000												
	1070	1040	1000													
	1030	990														
	990															
	X4															
TETRAHYDRON WITH		X-minimum = .125	X-maximum = .625													
		X1														
X4			X4													
980	970	950	930	930	950	970	980									
1030	1000	970	940	920	940	970	990									
1070	1030	980	970	940	970	990										
1100	1040	1030	1000	970	1000											
X2	1120	1110	1090	1050	1000	X3	Fig. (5B)									
	1100	1060	1050	1000												
	1070	1040	990													
	1030	990														
	980															
	X4															
TETRAHYDRON WITH		X-minimum = .25;	X-maximum = .25													
		X1														
X4			X4													
X4	X4															
X2	X3		Fig. (5C)													
	X4															

MAXIMUM VALUE YR = 8.88  
OPTIMUM COMBINATION: (5: 3: 0: 0)

Fig. 5. Computer output of ESP after harvesting of halophyte plant as affected by different combination gypsum, limestone, sulfuric acid and organic compost

NUMBER OF TREATMENTS = 19; NUMBER OF REPLICATES = 3  
EXPERIMENTAL DATA AND (MEAN)

1. 1.29	1.28	1.31 (1.29)	2. 1.29	1.26	1.34 (1.30)
3. 1.25	1.27	1.30 (1.27)	4. 1.28	1.24	1.25 (1.26)
5. 1.28	1.29	1.30 (1.29)	6. 1.29	1.24	1.22 (1.25)
7. 1.27	1.25	1.31 (1.28)	8. 1.25	1.32	1.24 (1.27)
9. 1.30	1.31	1.30 (1.30)	10. 1.22	1.28	1.31 (1.27)
11. 1.28	1.30	1.25 (1.28)	12. 1.25	1.26	1.25 (1.25)
13. 1.29	1.24	1.24 (1.26)	14. 1.25	1.32	1.35 (1.31)
15. 1.21	1.28	1.27 (1.25)	16. 1.33	1.22	1.25 (1.27)
17. 1.21	1.25	1.28 (1.25)	18. 1.28	1.25	1.24 (1.26)
19. 1.30	1.27	1.25 (1.27)			

GENERAL MEAN = 16.99754 MEAN ERROR = 0.4964015  
CORRELATION COEFFICIENT = 0.954934  
KERETTRION FISHER F(18,38) = 21.85232

REALITY ADEQUATE TEST OF THE MODEL THROUGH THE LAST FOUR TREATMENTS

EXPERIMENTAL TREAT.	TABULAR	DIFFERENCE	KERSTUDENT
16	1.27 - 1.27 =	-0.00	-0.04
17	1.25 - 1.25 =	0.00	0.82
18	1.26 - 1.27 =	-0.01	-0.72
19	1.27 - 1.28 =	-0.00	-0.20

THE SIGNIFICANT ADEQUATE SECTOR OF THE SQUARE TEST FOR THE CALCULATED VALUE, MULTIPLY BY 160

TETRAHYDRON WITH X<sub>1</sub> X<sub>4</sub>

X<sub>4</sub> 126 126 126 127 127 127 128 129 129 128 127 127 127 126 126 126  
126 126 126 126 126 127 127 127 129 128 128 127 127 127 127 126  
126 126 126 126 126 126 126 126 126 126 126 126 126 126 126 126  
127 126 126 126 125 125 126 127 128 128 128 127 127 127  
127 126 126 125 125 126 127 127 128 128 128 128 128  
128 127 126 125 125 126 127 127 127 128 128 128  
128 127 126 127 128 128 128 128 127 128 128  
129 128 128 129 129 129 129 128 127 128  
X<sub>2</sub> 130 130 130 131 130 130 129 128 127 X<sub>3</sub> Fig. (6A)

129 129 130 130 130 129 129 128  
128 128 128 128 128 128  
127 128 128 128 128  
127 127 127 127  
126 127 127  
126 126  
126  
X<sub>4</sub>

TETRAHYDRON WITH X<sub>1</sub> X<sub>4</sub>

X<sub>4</sub> 125 125 125 125 125 125 125 125 125  
125 125 125 125 127 127 127 127  
125 125 125 125 126 128 128 128  
125 125 126 128 129 130  
X<sub>2</sub> 126 127 129 130 131 X<sub>3</sub> Fig. (6B)

125 127 128 130  
125 127 128  
125  
X<sub>4</sub>

TETRAHYDRON WITH X<sub>1</sub> X<sub>4</sub>

X<sub>4</sub> X<sub>4</sub>  
129 Fig. (6C)  
X<sub>2</sub> X<sub>3</sub>  
X<sub>4</sub>

MINIMUM VALUE YR = 1.24  
OPTIMUM COMBINATION: (3: 3: 1: 1)

Fig. 6. Computer output of bulk density after harvesting of halophyte plant as affected by different combination gypsum, limestone, sulfuric acid and organic compost

NUMBER OF TREATMENTS = 19; NUMBER OF REPLICATES = 3  
EXPERIMENTAL DATA AND (MEAN)

1. 51.43	51.53	50.65 (51.28)	2. 51.43	52.43	49.34 (51.07)
3. 52.66	51.94	51.11 (51.90)	4. 51.56	53.15	52.66 (52.46)
5. 51.60	51.26	50.95 (51.27)	6. 51.44	53.07	53.97 (52.83)
7. 51.96	53.00	50.44 (51.81)	8. 52.75	50.01	53.35 (52.04)
9. 50.96	50.71	51.11 (50.93)	10. 53.86	51.00	50.69 (52.05)
11. 51.73	50.91	52.65 (51.76)	12. 52.97	52.36	52.66 (52.66)
13. 51.42	53.23	53.32 (52.66)	14. 52.97	50.24	49.07 (50.76)
15. 54.18	51.56	51.90 (52.35)	16. 49.86	54.04	52.90 (52.27)
17. 54.34	52.65	51.65 (52.88)	18. 51.59	52.88	53.38 (52.62)
19. 51.07	52.25	52.94 (52.09)			

GENERAL MEAN = 51.98913 MEAN ERROR = 0.5874369  
CORRELATION COEFFICIENT = 0.540446  
KERETTRION FISHER F(18,38) = 0.8710291

REALITY ADEQUATE TEST OF THE MODEL THROUGH THE LAST FOUR TREATMENTS

EXPERIMENTAL TREAT.	TABULAR	DIFFERENCE	KERSTUDENT
16	52.77 - 52.23 =	0.04	0.05
17	52.88 - 52.94 =	-0.06	-0.09
18	52.62 - 52.04 =	0.58	0.82
19	52.09 - 51.84 =	0.25	0.35

THE SIGNIFICANT ADEQUATE SECTOR OF THE SQUARE TEST FOR THE CALCULATED VALUE, MULTIPLY BY 10

TETRAHYDRON WITH X<sub>1</sub> X<sub>4</sub>

X<sub>4</sub> 525 524 523 522 520 519 517 515 512 515 517 519 520 522 523 524 525  
524 524 524 524 523 522 521 519 514 516 518 519 520 521 522 522  
523 524 525 525 525 525 524 520 516 517 518 519 520 520 520  
522 524 525 526 527 527 523 520 517 518 518 519 519 519  
521 523 525 527 525 522 520 518 518 518 518 518 518  
519 522 524 527 524 523 520 519 519 519 518 517  
516 520 524 521 519 518 518 518 519 518 517  
514 518 516 515 514 515 515 517 519 518  
X<sub>2</sub> 511 509 509 509 509 511 513 516 519 X<sub>3</sub> Fig. (7A)

514 512 511 511 512 513 515 518  
516 515 514 514 514 515 517  
519 517 516 516 516 517  
521 519 518 517 518  
522 520 519 519  
523 521 520  
524 522  
525  
X<sub>4</sub>

TETRAHYDRON WITH X<sub>1</sub> X<sub>4</sub>

X<sub>4</sub> 525 526 527 527 527 527 527 526 525  
526 528 529 530 522 522 521 520  
527 529 531 524 518 517 515  
527 530 523 518 513 511  
X<sub>2</sub> 527 521 516 511 508 X<sub>3</sub> Fig. (7B)

527 521 516 511  
527 521 515  
526 520  
525  
X<sub>4</sub>

TETRAHYDRON WITH X<sub>1</sub> X<sub>4</sub>

X<sub>4</sub> X<sub>4</sub>  
519 Fig. (7C)  
X<sub>2</sub> X<sub>3</sub>  
X<sub>4</sub>

MAXIMUM VALUE YR = 52.88  
OPTIMUM COMBINATION: (2.5: 2.5: 0.5: 2.5)

Fig. 7. Computer output of soil porosity [%] after harvesting of halophyte plant as affected by different combination gypsum, limestone, sulfuric acid and organic compost

shows the optimum combination for decreasing ESP was obtained by a rate the treatment of [5: 3: 0: 0] (of the 8 points score) [G, L, S and C], respectively. This reflects the marked affect of the [8.44, 2.18, 0.00 & 0.00] Mg $\text{fed}^{-1}$ , respectively. The general mean of ESP decreased from 38.01 to 12.40 to 8.80 from at the end of leaching process and after harvest, respectively.

#### **Bulk density and porosity**

Regarding bulk density "BD" and soil porosity "P" after harvesting data of Table 4 and Figs 6 and 7 shows a marked effect due to application of amendments. The optimum combination for decreasing BD and increasing P occurred with the mixture of G, L, S and C at the ratio of [3: 3: 1: 1], as well as [2.5, 2.5, 0.5, 2.5] (of the 8 points score), respectively.

#### **CONCLUSION**

It could be concluded that treating the saline soil with combination of gypsum, sulfuric acid, lime-stone, and organic compost followed by growing the halophyte plant *A. halimus* decreased EC, pH, SAR, ESP. Scanning the results illustrated in Table 5 show that the halophyte plant increased the effect of amendments. However no one

combination was most effective for all soil properties. To improve a specific character, it is easy to apply a mixture of gypsum, lime-stone, sulfuric acid and compost. The recommended combination to improve Sahl El-Tina saline sodic at the end of leaching process could be 1.0, 1.0, 2.5, and 3.5 (of the 8 points score) G, L, S and C, respectively i.e. the rate of 1.7, 0.73, 0.08, 3.6 Mg $\text{fed}^{-1}$  respectively. The recommended combination after harvesting of halophyte plant could be 2.0, 2.0, 3.0, 1.0 (of the 8 points score) i.e. 3.38, 1.45, 0.09 & 1.02 Mg $\text{fed}^{-1}$ , respectively.

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#### **REFERENCES**

- Abou Youssef, M.F. 2001. Use phosphogypsum fortified as a soil amendment for saline sodic soil in El-Salhiya plain. Zagazig J. Agric. Res., 28 (5): 889-911.
- Ahmad, M., B.H. Niazi and G.R. Sandhu. 1988. Effectiveness of



- gypsum, hydrogen chloride and organic matter for improvement of saline sodic soils. Pak. J. Agric. Res., 9 (3): 373-378.
- Al-Khateeb, S.A., A.M. Al-Gossibi. 2001. Performance of three halophytic species grown under saline condition. Zagazig J. Agric. Res., 28 (3):569:578.
- Anand, S. 1992. Effect of organic amendments on the nutrition and yield of wetland and sodic soil reclamation. J. Indian Soc. Soil Sci., 40(4):816-822.
- Avinelech, Y., M. Kochba, Y. Yotal, and D. Shkedi. 1990. On the use of municipal solid waste compost for the reclamation of saline and alkaline soils. Transaction 14th Inter. Cong. Of Soil Sci. Kyoto, Japan, 4:186-191.
- Baruah, T.C.; and H.P. Barthakur. 1997. A Textbook of Soil Analysis. Dept. Soil Sci., Assam Agric. Univ., Jorhat., India.
- Benz, L.C., F. Sandoval and W.O. Willis. 1967. Soil salinity changes with fallow and straw mulch on fallow. Soil Sci., 104:63-68.
- Bresler, E.; B.L. McNeal and D.L. Carter. 1982. Reclamation of saline and sodic soils. In: "Saline and sodic soils. Principles Dynamics-modeling". Springer-Verlag, Berling-Heidelberg, New York.
- Carter, D.L. and C.W. Robbins. 1978. Salt outflows from new and old irrigated land. Soil Sci. Soc. Am. J., 47:627-632.
- Chaganti, V. N. 2008. Reclamation potential of amendments for soils irrigated with saline-sodic drainage water. M.S. thesis, California State University, Fresno, USA.
- Chapman, H.D. and P.F. Partt. 1961. Methods of Analysis for Soils Plants and Waters. Div. Agric. Sci., Univ. of Calif. Berkely.
- Cheny, K. and R.S. Swift. 1984. The influence of organic matter on aggregate stability in some British soils. Soil Sci., 35:223-230.
- Chung, T.J. 2002. Computational fluid dynamics. Cambridge Univ.
- Diez, T. and M. Krauss. 1997. Effect of long-term compost application on yield and soil fertility. Agribiological Research - Zeitschrift Fur Agrarbiologie - AgriculturchemieOkologie, 50:78-84, Germany.
- Entry, J.A., B.H. Wood, J.H. Edwards, and C.W. Wood. 1997. Influence of organic by-products and nitrogen source on

- chemical and microbiological status of an agricultural soil. *Biol. Fertil. Soil*, 24:196-204.
- Gao, G. and C.C. hang. 1996. Charges in CEC and particle size distribution of soils associated with long term annual applications of cattle feedlot manure. *Soil Sci.*, 161:115-120.
- Giusquiani, P.L., M. Pagliai, G. Gigliotti, D. Bussinelli and A. Benetti. 1995. Urban waste compost: effects on physical, chemical and biochemical properties of soil. *Environ. Qual*, 24:175-182.
- Gupta I.C. 1990. Use of Saline Water in Agriculture. A study of arid and semi-arid zones in India. Revised edition. Oxford and IBH Publishing, New Delhi.
- Hanay, A. and N. Yardimci. 1992. A research on the effects of municipal compost and barnyard manure on some physical and chemical properties of soils and soil - water relations. *Doga-Transaction*]. *Agric. Forestry* 16:91-102, Ankara, Turkey.
- Ibrahim, S.M. and H. Shindo. 1999. Effect of continuous compost application on water - stable soil macroaggregation in a field subjected to double cropping. *Soil Sci. Plant Nutr.*, 45:1003-1007.
- Kamphorst, A.; and G.H. Aolt. 1976. Saline and sodic soils. In: Bolt, G.H., Bruggenuert, M.G.U., (eds), "Soil chemistry A Basic Elements". Elsevier, Amsterdam.
- Loveday, J. (1984). Amendments for reclaiming sodic soil. In Shainberg I., and Shalhevt, J., (eds) *Soil salinity under irrigation processes and management*. Springer-Verleg, Berlin, Heidelberg.
- Mamo, M., J.F. Moncrief, C.J. Rosen and T.R. Halbach. 2000. Municipal solid waste compost application on soil water and water stress in irrigated com. *Compost Sci. and Utiliz.*, 8:236-246.
- Mousa, K.F. and N.N. Youssef. 1992. Tetra factorial computer model for evaluating optimal agricultural parameters infertility studies. *Zagazig J. Agric. Res.*, 19: (6) 1992.
- Naeni, S.A. and H.F. Cook. 2000. Influence of municipal waste compost amendment on soil water and evaporation. *Commun. Soil Sci. Plant Anal.*, 31:3147-3161.
- Niazi, B.H., M. Ahmed, N. Hussain and M. Salim. 2001. Comparison of sand, gypsum

- and sulphuric acid to reclaim a Dense Saline Sodic Soil. *International J. of Agriculture and Biology*, 3: 316-318.
- Page, A.L. 1991. *Methods of Soil Analysis*, 2ND Edn., Am. Soc. Agron. and soil Sci. Am., Madison, Wisconsin, USA.
- Prather, R.J., J.O. Goertzen, J.D. Rhoades, and H. Frenkel. 1978. Efficient amendment use in sodic soil reclamation. *Soil Sci. Soc. Am. J.*, 42:782-786
- Prihar, S.S., S.K. Jalota and J. Lteiner. 1996. Residue management for reducing evaporation in relation to soil type and evaporability. *Soil Use and Mgt.*, 12:150-157.
- Sahin, U., O. Anapali, and A. Hanay. 2002. The effect of consecutive applications of leaching water applied in equal, increasing or decreasing quantities on soil hydraulic conductivity of a saline-sodic soil in the laboratory. *Soil Use and Mgt.*, 8:152-154.
- Singh, N.T.; G.S. Hira, and M.S. Bajwa. 1981. Use of amendments in reclamation of alkali soils in India. *Agrokemia es Talajtan*, 30: 158-177.
- Tanjii, K.K. 1990. *Agricultural salinity assessment and management*. Amer. Soc. Civil Eng. (ASCE) Manual No. 71. NY, USA.
- USDA 1954. *Diagnosis and improvement of saline and alkali soils*. Agriculture Hand Book No. 60 US Gov. Printing Office, Washington. USA.
- Wahid, A., S. Akhtar, L. Ali and E. Rasul. 1998. Amelioration of saline-sodic soils with organic matter and their use for wheat growth. *Commun. Soil Sci. Plant Anal.*, 29:2307-2318.
- Yahia, T.A., S. Miyamoto and J.L. Stroehlein. 1975. Effect of surface applied sulfuric acid on water penetration into dry calcareous and sodic soils. *Soil Sci. Soc. Am. proc.*, 39:1201-1204.

## الاستخدام الأمثل لتوليفات محسنات التربة لاستصلاح وإعادة تأهيل الأراضي المتأثرة بالأملاح

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أقيمت تجربة لتحديد التوليفة المثلى من الجبس والحجر الجيري وحامض الكبريتيك والكمبوست لاستصلاح الأراضي المتأثرة بالأملاح. وأجريت عملية الاستصلاح على مرحلتين حيث تم إمرار ٣٠ سم من الماء خلال عمود التربة وبعد الانتهاء تم زراعة نبات هالوفيت (*Atriplex halimus*) في كل عمود تربة مع إجراء عملية ري فقط دون نفاذ ماء من العمود. بعد الانتهاء من التجربة أدخلت النتائج إلى موديل الحاسب الآلي رباعي العوامل لتحديد التوليفة المثلى التي أدت إلى تحسين الصفات الكيميائية (EC, pH, SAR, ESP) والطبيعية (Bulk density and porosity). أوضحت النتائج ما يلي:

كل التوليفات من الجبس الزراعي والحجر الجيري وحامض الكبريتيك والكمبوست أدت إلى انخفاض EC, pH, SAR, ESP كما يمكن ملاحظة أن نبات القطف (*A. halimus*) أدى إلى زيادة فعالية محسنات التربة المضافة.

التوليفة المثلى التي أدت إلى خفض درجة التوصيل الكهربائي من ٢٨,٨ إلى ٥,٨ ديسيمنز / م من الجبس، الحجر الجيري، حامض الكبريتيك والكمبوست بعد انتهاء عملية الغسيل هي ١,٦٩، ١,٤٥، ٠,٠٩، ٢,٠٣ طن / فدان على الترتيب وهي نفس التوليفة التي أظهرت انخفاض ملحوظ بعد حصاد نبات القطف حيث خفضت درجة التوصيل الكهربائي من ٥,٨ إلى ٢,١ ديسيمنز / م. بينما التوليفة المثلى التي أدت إلى خفض حموضة التربة من ٨,٤ إلى ٧,٩٩ بعد انتهاء عملية الغسيل هي صفر، صفر، صفر، ٨,١٢ طن / فدان على الترتيب بينما التوليفة التي أظهرت انخفاض ملحوظ في حموضة التربة بعد حصاد نبات القطف هي صفر، صفر، ٠,٢٥ صفر طن / فدان حيث خفضت درجة تفاعل التربة من ٨,١ إلى ٧,٧.

التوليفة المثلي التي أدت إلى خفض SAR من ١٢,٩٠ إلى ٦,٠٥ من الجبس، الحجر الجيري، حامض الكبريتيك والكمبوست بعد انتهاء عملية الغسيل هي ١,٦٩، ٠,٧٣، ٠,١٣، ٢,٠٣ طن / فدان علي الترتيب بينما التوليفة التي أظهرت انخفاض ملحوظ في SAR بعد حصاد نبات القطف هي ١,٦٩، ١,٤٥، ٠,٠٩، ٢,٠٣ طن / فدان حيث خفضت SRA إلى ٤,٠٦. بينما التوليفة المثلي التي أدت إلى خفض حموضة التربة من ٨,٤ إلى ٧,٩٩ بعد انتهاء عملية الغسيل هي صفر، صفر، صفر، ٨,١٢ طن / فدان علي الترتيب بينما التوليفة التي أظهرت انخفاض ملحوظ في حموضة التربة بعد حصاد نبات القطف هي صفر، صفر، ٠,٢٥، صفر طن / فدان حيث خفضت درجة تفاعل التربة من ٨,١ إلى ٧,٧. بينما التوليفة المثلي التي أدت إلى خفض ESP بعد انتهاء عملية الغسيل هي ٣,٣٨، ٠,٧٣، ٠,٠٩، ٢,٠٣ طن / فدان علي الترتيب حيث خفضت ESP من ٣٨,٠١ إلى ١٢,٤ بينما التوليفة التي أظهرت انخفاض ملحوظ في ESP بعد حصاد نبات القطف هي ٨,٤٤، ٢,١٨، صفر، صفر طن / فدان حيث خفضت ESP إلى ٨,٨٠.